phase column eluted with water-MeOH, 95:5. All velocities were determined from at least four time points that were within the initial reaction rate period.

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# Synthesis and Oral Antiallergic Activity of Carboxylic Acids Derived from Imidazo[2,1-c ][1,4]benzoxazines, Imidazo[1,2-a ]quinolines, Imidazo[1,2-a ]quinoxalines, Imidazo[1,2-a ]quinoxalinones, Pyrrolo[1,2-a ]quinoxalinones, Pyrrolo[2,3-a ]quinoxalinones, and Imidazo[2,1-b]benzothiazoles 

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$4 H$-Imidazo $[2,1-c][1,4]$ benzoxazine-2-carboxylic acid (3) was found to possess potent activity in the IgE-induced rat passive cutaneous anaphylaxis model which may be predictive of clinical antiallergic activity. Compared to disodium cromoglycate (DSCG, 1), 3 was less active following iv administration but unlike DSCG showed very significant oral activity. To explore the structural requirements for this activity, a range of tricyclic compounds was prepared and their activities were measured. Individual 2 -carboxylic acids derived from imidazo $1,2-a]$ quinolines, imidazo $[1,2-a]$ quinoxalines, imidazo $[1,2-a]$ quinoxalinones, pyrrolo $[1,2-a$ ]quinoxalinones, pyrrolo[ $2,3-a]$ quinoxalinones, and imidazo $[2,1-b]$ benzothiazoles showed iv activities up to $10^{3}$ times as potent as DSCG and many of them showed significant oral activity. From these, imidazo[1,2-a]quinoxaline-2-carboxylic acid 114 has been chosen for further development.

Asthma is a disease of uncertain etiology primarily involving the small bronchi and manifested clinically by intermittent wheezing and dyspnea of varying intensity. ${ }^{1}$ Treatment for the condition has involved use of bronchodilators, e.g., $\beta$-adrenergic agonists and theophylline and corticosteroids, but a major step forward was made with the introduction of disodium cromoglycate (DSCG, $1)^{2}$ as a prophylactic agent against the disease. Subsequent clinical trials have shown the efficacy of DSCG in suitable patients, ${ }^{3,4}$ but it is not active orally and has to be insufflated as a powder. The discovery of an oral, prophylactic antiasthmatic agent remains a goal of a number of laboratories, ${ }^{5}$ and we have previously reported one such compound, ${ }^{6,7} 2$ (RU 31156, Sudexanox ${ }^{8}$ ).


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[^0]The origin of the work described in this paper was the unexpected rearrangement which gave 4 H -imidazo[2,1c] $[1,4]$ benzoxazines as previously described. ${ }^{9}$ ( $4 H$ -Imidazo[2,1-c][1,4]benzoxazine-2-carboxylic acid (3) was found to possess significant activity in the rat IgE -mediated passive cutaneous anaphylaxis test $\left(\mathrm{ED}_{50}=2.89\right.$ $\mathrm{mg} / \mathrm{kg}$ iv, Table I), a possible but not unequivocally predictive model for clinical efficacy. ${ }^{5}$ This result led us to undertake a more systematic examination of structureactivity requirements for PCA activity in related tricyclic heterocyclic systems. Initial approaches retained the imidazole ring of 3 and varied the 4,5 -positions to give series $4 a-e, 4 h, 5$, and 6 , and then retaining the quinoxalin-4( 5 H )-one system, the five-membered ring was varied to give pyrrole 4f, pyrazole 4g, and triazole 4i. Subsequent work produced ring systems 7-10 in order to observe the effects of a wider variety of modifications.

## Chemistry

Structures $4 a-$ j and 5-10 show the variety of ring systems synthesized (Chart I).
(a) Imidazo[2,1-c][1,4]benzoxazines (3, 11-39; Table I). The basic method of preparation of this system was described earlier, ${ }^{9}$ and the derivatives studied are listed in Table I along with their pharmacological properties.
(b) Imidazo[1,2-a ]quinolines (46-113; Tables II and III). Treatment of the quinolinium quaternary salt 40 with ammonium acetate in glacial acetic acid heated under reflux was reported ${ }^{11}$ to give the imidazolidine 41, but the

[^1]Scheme I

structure was subsequently amended ${ }^{12,13}$ to $\mathbf{4 2}$, formed by double bond migration. We reasoned that, by using ethyl bromopyruvate instead of phenacyl bromide as the quaternizing agent to give 43, then the equivalent reaction should give the dihydroimidazo[1,2-a]quinoline ester 44. The product isolated from this reaction was a mixture of esters 44 and 45 , which were separated by chromatography and hydrolyzed to the respective acids 46 and 47. The use of hydroxylamine hydrochloride ${ }^{14}$ instead of ammonium acetate as the nitrogen source in the cyclization step gave the ester 45 free from 44 but only in low yield. Since both acids 46 and 47 were more active in the rat PCA screen than 3 , but 46 was much less active po, it was decided to concentrate on the series derived from ester 45; therefore an improved synthetic route was required. This was achieved by reacting 2 -aminoquinoline with ethyl bromopyruvate, giving a product which on heating in ethanol followed by basification yielded a compound identical in all respects with ester 45. That the product from both routes is 45 confirms that quaternization of 2 -aminoquinoline takes place on the 1-nitrogen prior to cyclization. An analogous preparation of imidazopyridines from 2aminopyridine has been reported. ${ }^{15}$



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(10) Goose, J.; Blair, A. M. J. N. Immunology 1969, 16, 749.
(11) Krohnke, K.; Zecher, W. Chem. Ber. 1962, 95, 1128. Krohnke, K. Angew. Chem., Int. Ed. Engl. 1963, 2, 225.
(12) Cookson, R. F.; Nowotnik, D. P.; Parfitt, R. T. J. Chem. Soc., Chem. Commun. 1974, 911. Airey, J. E.; Cookson, R. F.; Kende, A. S.; Nowotnik, D. P.; Parfitt, R. T. J. Chem. Soc., Perkin Trans. 1 1976, 201.
(13) Habermalz, U.; Krohnke, F.; Reinshagen, B. Chem. Ber. 1975, $108,984$.
(14) Sale, A. O.; Toja, E.; Galliani, G.; Lemer, L. J. U. K. Patent 1484 615, 23 Nov. 1974.
(15) Lombordino, J, G. J. Org. Chem. 1965, 30, 2403. Smakula Hand, E.; Paudler, W. W. Tetrahedron 1982, 38, 49.

Scheme II $^{a}$

${ }^{a}$ Reagents: (a) (i) $\mathrm{NaOH} ; \mathrm{H}_{2} \mathrm{O}, \mathrm{EtOH}$, (ii) HCl ; (b) $\mathrm{CDI}, \mathrm{H}_{2} \stackrel{\wedge}{\mathrm{~N}}$ $\overline{\mathrm{C}=\mathrm{N}-\mathrm{N}=\mathrm{NH}}$; (c) $\mathrm{SOCl}_{2}$; (d) $\mathrm{NEt}_{3}$, glycerol acetonide; $\mathrm{ClCH}_{2} \mathrm{C}$ $\mathrm{H}_{2} \mathrm{Cl}$; (e) citric acid; $\mathrm{H}_{2} \mathrm{O}$; (f) $\mathrm{Na}, \mathrm{NH}_{3}$; EtOH ; (g) $\mathrm{LiBH}_{4}$; THF; (h) $\mathrm{MnO}_{2} ; \mathrm{CHCl}_{3} ;$ (i) $\mathrm{NH}_{2} \mathrm{OH} \cdot \mathrm{HCl}, \mathrm{NaOAc} ; \mathrm{H}_{2} \mathrm{O}, \mathrm{EtOH} ;(\mathrm{j}) \mathrm{H}_{2}, \mathrm{Pd} / \mathrm{C}$; $\mathrm{MeOH}, \mathrm{HCl}$; (k) $\mathrm{Ac}_{2} \mathrm{O}$; (l) $\mathrm{NaN}_{3}, \mathrm{NH}_{4} \mathrm{Cl}$; DMF.

## Scheme III $^{a}$


${ }^{a}$ Reagents: (a) $\mathrm{LiAlH}_{4}$; THF; (b) (i) $\mathrm{BrCH}_{2} \mathrm{COCO}_{2} \mathrm{Et}$; DME, (ii) $\Delta$; EtOH ; (c) (i) $\mathrm{NaOH} ; \mathrm{H}_{2} \mathrm{O}, \mathrm{EtOH}$, (ii) HCl ; (d) (i) $\mathrm{MnO}_{2} ; \mathrm{CHCl}_{3}$, (ii) $\mathrm{NaOAc}, \mathrm{NH}_{2} \mathrm{OH} \cdot \mathrm{HCl} ; \mathrm{H}_{2} \mathrm{O}, \mathrm{EtOH}$, (iii) $\mathrm{Ac}_{2} \mathrm{O}$; (e) $\mathrm{Br}_{2} ; \mathrm{AcONa}$, AcOH ; (f) (i) MeCHBrCOCO 2 Et ; DME, (ii) $\Delta$; EtOH.

Chart I


A variety of substituted 2 -aminoquinolines were prepared and subjected to the same quaternization process (Scheme I). Substituents on the ring nucleus were also modified (Schemes II and III and the Experimental Section).
(c) Imidazo[1,2-a ]quinoxalines (114-122; Table IV). The reaction of 2 -aminoquinoxalines with ethyl bromopyruvate as in the preparation of $\mathbf{4 5}$ followed by modification of the ester group (Scheme IV) gave a series of imidazo $[1,2-a]$ quinoxalines.
(d) Imidazo[1,2-a ]quinoxalin-4(5H)-ones (123-136; Table V). The reaction of 2 -amino- 3 -chloroquinoxaline or 2 -amino-3-methoxyquinoxaline ${ }^{16}$ with ethyl bromopyruvate was found to be accompanied by hydrolysis of the 3 -substituent to give the imidazo[1,2-a]quinoxalin-4( 5 H )-one ester 123. The 5 -nitrogen atom was readily alkylated, and the resulting esters were then hydrolyzed to the corresponding acids (Scheme V). Modification of the
(16) Cheeseman, G. W. H. J. Chem. Soc. 1955, 1804.

Scheme IV ${ }^{\text {a }}$

${ }^{a}$ Reagents: (a) (i) $\mathrm{BrCH}_{2} \mathrm{COCO}_{2} \mathrm{Et}$; DME, (ii) $\Delta$; EtOH ; (b) (i) $\mathrm{NaOH} ; \mathrm{H}_{2} \mathrm{O}, \mathrm{EtOH}$, (ii) HCl ; (c) $\mathrm{LiBH}_{4}$; THF; (d) $\mathrm{CDI}, \mathrm{NH}_{2} \mathrm{C}=$ $\mathrm{N}-\mathrm{N}=\mathrm{N}-\mathrm{NH} ; \mathrm{DMF}$; (e) $\mathrm{MnO}_{2} ; \mathrm{CHCl}_{3}$, room temperature; (f) $\mathrm{MnO}_{2} ; \mathrm{CHCl}_{3}, \Delta ;$ (g) (i) $\mathrm{NH}_{2} \mathrm{OH} \cdot \mathrm{HCl}, \mathrm{NaOAc} ; \mathrm{EtOH}, \mathrm{H}_{2} \mathrm{O}$, (ii) $\mathrm{Ac}_{2} \mathrm{O}$, (iii) $\mathrm{NaN}_{3}, \mathrm{NH}_{4} \mathrm{Cl} ; \mathrm{DMF}$.

Scheme $\mathbf{V}^{a}$

${ }^{a}$ Reagents: (a) $\mathrm{BrCH}_{2} \mathrm{COCO}_{2} \mathrm{Et}$; DME; (b) $\Delta$; EtOH ; (c) NaH , RI; DMF; (d) (i) $\mathrm{NaOH} ; \mathrm{H}_{2} \mathrm{O}, \mathrm{EtOH}$, (ii) HCl .

## Scheme VI ${ }^{a}$


${ }^{a}$ Reagents: (a) (i) $\mathrm{BrCH}_{2} \mathrm{COCO}_{2} \mathrm{Et}$; DME, (ii) $\Delta$, EtOH ; (b) (i) $\mathrm{NaOH} ; \mathrm{H}_{2} \mathrm{O}, \mathrm{EtOH}$, (ii) HCl .
ester group of 124 yielded derivatives 134-136.
(e) Imidazo[1,2-a ]quinazolin-5(4H)-ones (137,138; Table VI). The reaction of 2 -amino- 3 -ethylquinazolin$4(3 \mathrm{H})$-one with ethyl bromopyruvate as in the preparation of 45 gave the ester 137 , which was hydrolyzed by base to the acid 138 (Scheme VI).

Scheme VII ${ }^{a}$

${ }^{a}$ Reagents: (a) $\mathrm{MeO}_{2} \mathrm{CC} \equiv \mathrm{CCO}_{2} \mathrm{Me}, \mathrm{Et}_{3} \mathrm{~N}$; DMF; (b) $\mathrm{HC} \equiv$ $\mathrm{CCO}_{2} \mathrm{Et}, \mathrm{Et}_{3} \mathrm{~N}$; DMF; (c) (i) $\mathrm{NaOH} ; \mathrm{H}_{2} \mathrm{O}, \mathrm{EtOH}$, (ii) HCl .

Scheme VIII ${ }^{a}$

${ }^{a}$ Reagents: (a) (i) $\mathrm{KMnO}_{4}$; aqueous NaOH , (ii) concentrated HCl ; (b) (i) $\mathrm{PtO}_{2}+5 \% \mathrm{Pd}-\mathrm{C}, \mathrm{H}_{2}$; aqueous NaOH , (ii) concentrated HCl ; (c) MeOH , dry HCl ; (d) RI , NaH ; DMF; (e) (i) NaOH ; $\mathrm{H}_{2} \mathrm{O}, \mathrm{EtOH}$, (ii) concentrated HCl .
(f) Pyrrolo[1,2-a ]quinoxolin-4(5H)-ones (141-163; Table VII). The reaction of the benzimidazole quaternary salt 139 with dimethyl acetylene dicarboxylate was shown ${ }^{17}$ to give the pyrrolo[1,2-a]quinoxalin-4(5H)-one diester 141. We found that the use of methyl propiolate gave the corresponding monoester 142 (Scheme VII). With use of this method, a series of acids, $145-157$, were prepared, and then a number of 5 -ethyl derivatives, 158-163, were synthesized for comparison.
(g) Pyrazolo[2,3-a ]quinoxalin-4(5H)-ones (166-172; Table VIII). An analogous series of pyrazolo[2,3-a]-quinoxalin- $4(5 H)$-one acids and esters were prepared as

## Scheme IX ${ }^{a}$


${ }^{a}$ Reagents: (a) $\mathrm{RC}\left(\mathrm{OEt}_{3}\right.$; (b) (i) $\mathrm{NaOH} ; \mathrm{H}_{2} \mathrm{O}, \mathrm{EtOH}$, (ii) HCl ; (c) RCOCl ; (d) $\left(\mathrm{CH}_{3}\right)_{2} \mathrm{CHCH}_{2} \mathrm{CH}_{2} \mathrm{ONO}$.
shown in Scheme VIII following literature precedents. ${ }^{18}$ Thus (o-nitrophenyl)hydrazine was condensed with ethyl acetopyruvate ${ }^{19}$ to give, after hydrolysis, the pyrazole acid 164. Oxidation of the pyrazole methyl group using potassium permanganate gave the diacid 165, catalytic reduction of the nitro group of which converted it to the cyclized acid amide 166.

Esterification of 166 gave the ester 167 , which was alkylated on nitrogen to produce esters 168 and 169, hydrolysis of which gave the acids 170 and 171 . Acid 172 was prepared directly from 167 by alkylation and in situ hydrolysis.
(h) Imidazo[1,5-a ]quinoxalinones, Triazolo[1,5-a]quinoxalinones, and Triazolo[1,5-a ]benzoxazinones (Scheme IX; Table IX). A number of compounds were also prepared in related ring systems containing a 3 -position (rather than 2-position) carboxyl group. Treatment of the diamine ${ }^{9,20} 173$ with triethyl orthoformate or triethyl orthoacetate gave the imidazo[1,5-a]quinoxalinone esters 174 and 175 , respectively. Reaction of 173 with acid chlorides gave the amides 176 and 177. Hydrolysis of 174 and 175 gave the acids 178 and 179 , respectively. Similar base treatment of 176 and 177 caused cyclization followed by hydrolysis to give the respective acids 180 and 181. The diamine ${ }^{9,21} 182$ was treated with isopentyl nitrite to give the triazole ester 183, and diamine 173 likewise yielded 184,

[^2]Table I. $4 H$-Imidazo[2,1-c][1,4]benzoxazines


| compd | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | $\mathrm{R}^{4}$ | $\mathrm{R}^{8}$ | preparation ${ }^{\text {d }}$ | antiallergic activity, IgE, rat PCA:$\mathrm{ED}_{50},{ }^{6} \mathrm{mg} / \mathrm{kg}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  | iv | po |
| 1 |  | DSCG |  |  |  | 1.21 (1.04-1.42) | inactive |
| 2 |  | Sudexanox |  |  |  | 0.005 (0.004-0.006) | 0.19 (0.07-0.030) |
| 3 | H | $\mathrm{CO}_{2} \mathrm{H}$ | H | H | A | 2.89 (2.22-3.63) | 3.20 (2.47-3.45) |
| 11 | H | $\mathrm{CO}_{2} \mathrm{Me}$ | H | H | A |  | 10.8 (7.73-15.31) |
| 12 | H | $\mathrm{CH}_{2} \mathrm{OH}$ | H | H | A |  | 2.73 (0.29-6.49) |
| 13 | H | CHO | H | H | A | 4.79 (1.73-30.03) | 4.27 (2.71-7.46) |
| 14 | H | $\mathrm{CH}=\mathrm{NOH}$ | H | H | A |  | inactive $^{\text {b }}$ |
| 15 | H | CN | H | H | A |  | inactive ${ }^{\text {c }}$ |
| 16 | H |  | H | H | A | 0.33 (0.21-0.51) | inactive ${ }^{\text {c }}$ |
| 17 | H | $\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{H}$ | H | H | A | inactive ${ }^{\text {c }}$ |  |
| 18 | H |  | H | H | A | inactive $^{\text {c }}$ |  |
| 19 | $\mathrm{CO}_{2} \mathrm{H}$ | $\mathrm{CO}_{2} \mathrm{Me}$ | H | H | A | inactive ${ }^{\text {c }}$ |  |
| 20 | $\mathrm{CO}_{2} \mathrm{Et}$ | $\mathrm{CO}_{2} \mathrm{Me}$ | H | H | A | inactive ${ }^{\text {c }}$ |  |
| 21 | $\mathrm{CO}_{2} \mathrm{H}$ | $\mathrm{CO}_{2} \mathrm{H}$ | H | H | A | inactive ${ }^{\text {c }}$ |  |
| 22 |  | $\mathrm{CO}_{2} \mathrm{H}$ | H | H | A | inactive ${ }^{\text {c }}$ |  |
| 23 | $\mathrm{CO}_{2} \mathrm{H}$ | H | H | H | A | inactive ${ }^{\text {c }}$ |  |
| 24 | H |  | H | H | B | 0.58 (0.43-0.74) | inactive ${ }^{\text {c }}$ |
| 25 | H |  | H | H | B | 0.33 (0.12-1.07) | inactive $^{\text {c }}$ |
| 26 | H |  | H | H | B | inactive ${ }^{\text {c }}$ |  |
| 27 | $\mathrm{CH}_{2} \mathrm{OH}$ | $\mathrm{CO}_{2} \mathrm{Me}$ | H | H | B |  | inactive ${ }^{\text {c }}$ |
| 28 | $\mathrm{CH}_{2} \mathrm{OH}$ | $\mathrm{CO}_{2} \mathrm{H}$ | H | H | B |  | inactive ${ }^{c}$ |
| 29 | H | $\mathrm{CO}_{2} \mathrm{H}$ | H | $\mathrm{NO}_{2}$ | A | 2.22 (1.31-4.63) |  |
| 30 | H | $\mathrm{CO}_{2} \mathrm{Me}$ | H | NHAc | A |  | 3.48 (2.89-3.99) |
| 31 | H | $\mathrm{CO}_{2} \mathrm{H}$ | H | NHAc | A | 0.073 (0.047-0.108) | inactive |
| 32 | H | $\mathrm{CO}_{2} \mathrm{Me}$ | H | $\mathrm{NHCOCO}_{2} \mathrm{Et}$ | A |  | 0.50 (0.37-0.65) |
| 33 | H | $\mathrm{CO}_{2} \mathrm{H}$ | H | $\mathrm{NHCOCO}_{2} \mathrm{H}$ | A | 0.0172 (0.0103-0.0214) | 3.85 (2.71-5.46) |
| 34 | H | $\mathrm{CO}_{2} \mathrm{H}$ | H | Cl | B | 0.42 (0.34-0.44) | 0.13 (0.04-0.45) |
| 35 | H | $\mathrm{CO}_{2} \mathrm{Me}$ | H | Cl | B |  | 2.79 (1.75-4.06) |
| 36 | H | $\mathrm{CO}_{2} \mathrm{Me}$ | H | Me | B |  | 1.46 (0.66-3.33) |
| 37 | H | $\mathrm{CO}_{2} \mathrm{H}$ | H | Me | B | 0.35 (0.26-0.50) | 0.37 (0.11-1.43) |
| 38 | H | $\mathrm{CO}_{2} \mathrm{H}$ | Me | H | B | 1.76 (0.91-3.86) | 8.46 (4.38-21.4) |
| 39 | H | $\mathrm{CO}_{2} \mathrm{H}$ | Et | H | B | 2.6 (1.61-7.0) |  |

${ }^{a} 95 \%$ confidence limits in parentheses. ${ }^{b} 42 \%$ inhibition at $10 \mathrm{mg} / \mathrm{kg}$. ${ }^{c}$ Inactive at 1 and $10 \mathrm{mg} / \mathrm{kg}$. ${ }^{d} \mathrm{~A}$, ref 9 ; B, see the Experimental Section.
hydrolysis of which produced the acid 185.
(i) Miscellaneous Ring Systems (Table X). Schemes X and XI show the routes used to synthesize imidazo-[1,2-a] pyridine ${ }^{22}$ 186, pyrrolo[1,2-a]quinolines 187 and 188, imidazo[2,1-a] isoquinolines 189 and $190,{ }^{23}$ thieno[3,2-a]naphthalene $191,{ }^{24}$ imidazo[2,1-c] benzothiadiazines 194-197, and imidazo[2,1-b]benzothiazoles 198-205. ${ }^{25}$
(22) Lombardino, J. G. J. Org. Chem. 1965, 30, 2403.
(23) Kuzmenko, T. A.; Kuzmenko, V. V.; Simonov, A. M.; Simkin, B. Y. Khim. Geterotsikl. Soedin. 1980, 12, 1656.
(24) Campaigne, E.; Cline, R. E. J. Org. Chem. 1956, 21, 32. Grunhaus, H.; Pailer, M.; Stof, S. J. Heterocycl. Chem. 1976, 13, 1161.
(25) Abignente, E.; Arena, F.; De Caprariis, P.; Parent, L. Farmaco, Ed. Sci. 1977, 32, 735.

## Biological Results and Discussion

All compounds were tested for their ability to inhibit the IgE-mediated passive cutaneous anaphylaxis (PCA) reaction in rats passively sensitized to ovalbumen. ${ }^{6,7,10}$
(a) Imidazo[1,2-c ][1,4]benzoxazines. Table I shows that significant PCA activity resides in the acid 3 by both the iv and po dosing schedules. The latter result is of course of great interest in view of the lack of oral activity of DSCG. Further, the level of oral activity is surprisingly close to that of the iv result. Significant activity is shown by 2 -substituents which could be metabolized to the acid (e.g., 11-13, 24, 25) or which mimic this group (e.g., 16). The presence of 1 -substituents or even the movement of the acid group to this position (23) completely abolishes activity. Activity is enhanced by the presence of both

Table II. (Benzo unsubstituted)imidazo[1,2-a]quinoline-2-carboxylic Acids and Derivatives



Table III. (Benzo substituted)imidazo[1,2-a]quinoline-2-carboxylic Acids and Derivatives


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|  |  |  |  |  |  |  | recrystn | yield, |  |  | antiallergic activity, IgE, rat PCA:$E D_{\text {so }}, \mathrm{mg} / \mathrm{kg}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| compd | $\mathrm{R}^{8}$ | $\mathrm{R}^{7}$ | $\mathrm{R}^{6}$ | $\mathrm{R}^{5}$ | $\mathrm{R}^{2}$ | mp, ${ }^{\circ} \mathrm{C}$ | solvent | \% | formula | anal. | iv | po |
| 79 | Cl | H | H | H | $\mathrm{CO}_{2} \mathrm{Et}$ | 221-222 | $\mathrm{CHCl}_{3} / \mathrm{Et}_{2} \mathrm{O}$ | 50 | $\mathrm{C}_{14} \mathrm{H}_{11} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Cl}$ | C,H,N,Cl |  | 1.72 (0.97-3.45) |
| 80 | Cl | H | H | H | $\mathrm{CO}_{2} \mathrm{H}$ | 288-289 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 64 | $\mathrm{C}_{12} \mathrm{H}_{7} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Cl}$ | C,H,N,Cl | 0.069 (0.046-0.095) | $0.11(0.091-0.135)$ |
| 81 | MeO | H | H | H | $\mathrm{CO}_{2} \mathrm{H}$ | 272-273 | $\mathrm{AcOH}{ }^{\text {a }}$ | 63 | $\mathrm{C}_{13} \mathrm{H}_{40} \mathrm{~N}_{2} \mathrm{O}_{3}$ | C,H,N | 0.038 (0.026-0.058) | $0.14(0.07-0.30)$ |
| 82 | Et | H | H | H | $\mathrm{CO}_{2} \mathrm{H}$ | 241-244 | EtOH | 72 | $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}$ | C,H,N | 0.16 (0.08-0.37) |  |
| 83 | SMe | H | H | H | $\mathrm{CO}_{2} \mathrm{H}$ | 262-264 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 81 | $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S} \cdot \mathrm{H}_{2} \mathrm{O}$ | C,H,N,S | 0.095 (0.058-0.15) |  |
| 85 | SOMe | H | H | H | $\mathrm{CO}_{2} \mathrm{H}$ | 261-263 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 86 | $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$ | C,H,N,S | 0.076 (0.043-0.12) |  |
| 87 | $\mathrm{SO}_{2} \mathrm{Me}$ | H | H | H | $\mathrm{CO}_{2} \mathrm{H}$ | 261-265 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 81 | $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{~S}$ | C,H,N,S | 0.028 (0.021-0.03) | inactive ${ }^{\text {b }}$ |
| 88 | $\mathrm{SO}(=\mathrm{NH}) \mathrm{Me}$ | H | H | H | $\mathrm{CO}_{2} \mathrm{H}$ | 235-237 | EtOH/H2O | 76 | $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}-0.5 \mathrm{H}_{2} \mathrm{O}$ | C,H,N,S | 0.095 (0.058-0.15) |  |
| 89 |  |  |  |  |  | 290-292 | EtOH/H2O | 84 | $\mathrm{C}_{13} \mathrm{H}_{9} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Cl} \cdot \mathrm{H}_{2} \mathrm{O}$ | C,H,N,Cl | 1.99 (1.14-3.89) |  |
| 90 | $\mathrm{CF}_{3}$ | H | H | $\stackrel{\mathrm{H}}{ }$ | $\mathrm{CO}_{2} \mathrm{H}$ | 264-266 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 91 | $\mathrm{C}_{13} \mathrm{H}_{7} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~F}_{3}$ | C,H,N | 0.070 (0.04-0.154) |  |
| 91 | MeO | H | H | H | $\mathrm{CH}_{2} \mathrm{OH}$ | 173-175 | THF/ $\mathrm{H}_{2} \mathrm{O}$ | 89 | $\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}$ | C,H,N | 0.27 (0.18-0.36) | 0.24 (0.03-22.6) |
| 92 | MeO | H | H | ${ }^{\mathrm{H}}$ | CHO | 190-193 | EtOAc/Et ${ }_{2} \mathrm{O}$ | 80 | $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{2}$ | C,H,N | 0.12 (0.05-0.37) | 0.40 (0.24-0.76) |
| 93 | MeO | H | H | H | $c_{N_{N}^{N-N}}^{N-N}$ | 305-306 | DMSO | 88 | $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{~N}_{6} \mathrm{O}$ | C,H,N | 0.01 (0.003-0.028) | 0.12 (0.002-24.1) |
| 94 | MeO | H |  |  | $\mathrm{CONH}_{2}$ | 273-276 |  | 40 |  | C,H,N |  | inactive ${ }^{\text {c }}$ |
| 95 | MeO | H | H | H | $\mathrm{CH}_{2} \mathrm{NH}_{2} \cdot 2 \mathrm{HCl}$ | 274-277 | $\mathrm{MeOH} / \mathrm{Et}_{2} \mathrm{O}$ | 74 | $\mathrm{C}_{13} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O} \cdot 2 \mathrm{HCl} \cdot 1.5 \mathrm{H}_{2} \mathrm{O}$ | $\mathrm{C}, \mathrm{H}, \mathrm{~N}, \mathrm{Cl}$ | $3.39(2.22-5.31)$ |  |
| 96 | MeO | H | H | H |  | 298-301 | DMF/EtOAc | 95 | $\mathrm{C}_{14} \mathrm{H}_{11} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ | $\mathbf{C , H , N}$ | inactive ${ }^{c}$ |  |
| 97 | H | Cl | H | H | $\mathrm{CO}_{2} \mathrm{H}$ | 278-279 | EtOH/H2O | 93 | $\mathrm{C}_{12} \mathrm{H}_{7} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Cl} \cdot \mathrm{H}_{2} \mathrm{O}$ | C,H,N,Cl | 0.29 (0.22-0.36) |  |
| 98 | $\stackrel{\mathrm{H}}{\mathrm{H}}$ | $\mathrm{NO}_{2}$ | $\stackrel{H}{H}$ | H | $\mathrm{CO}_{2} \mathrm{H}$ | 311-313 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 90 | $\mathrm{C}_{12} \mathrm{C}_{2} \mathrm{~N}_{3} \mathrm{O}_{4}$ | C,H,N | $0.28(0.20-0.35)$ |  |
| 99 | H | MeO | H | H | $\mathrm{CO}_{2} \mathrm{H}$ | 265-266 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 79 | $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{3}$ | C,H,N | 0.03 (0.003-0.48) |  |
| 100 | $\stackrel{\mathrm{Cl}}{\mathrm{Cl}}$ | H H | H | $\stackrel{\mathrm{Cl}}{\mathrm{Cl}}$ | $\mathrm{CO}_{2} \mathrm{Et}$ $\mathrm{CH}_{2} \mathrm{OH}$ | 229-230 | $\mathrm{CHCl}_{3} / \mathrm{Et}_{2} \mathrm{O}$ | 79 47 | ${ }_{\mathbf{C}}^{\mathrm{C}_{14} \mathrm{H}_{1} \mathrm{H}_{2} \mathrm{~N}_{2} \mathrm{OCCl}_{2} \mathrm{Cl}_{2}}$ | $\mathrm{C}, \mathrm{H}, \mathrm{N}, \mathrm{Cl}$ |  | 1.49 (0.30-10.04) |
| 101 | $\stackrel{\mathrm{Cl}}{\mathrm{Cl}}$ | H <br> H | H <br> H | Cl Cl | $\xrightarrow{\text { CHO }}$ | 207-210 | $\underset{\mathrm{CHCl}_{3} / \mathrm{Et}_{2} \mathrm{O}}{\mathrm{EtOAc} / \mathrm{Et}_{2} \mathrm{O}}$ | 47 50 | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{OCl}_{2}$ $\mathrm{C}_{12} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{OCl}_{2}$ | $\begin{aligned} & \mathrm{C}, \mathrm{H}, \mathrm{~N}, \mathrm{Cl} \\ & \mathrm{C}, \mathrm{H}, \mathrm{~N}, \mathrm{Cl} \end{aligned}$ | 0.16 (0.02-0.38) | 0.5(0.21-7.6) |
| 102 | CI | H | H | Cl | CHO | 280-285 | $\mathrm{CHCl}_{3} / \mathrm{Et}_{2} \mathrm{O}$ | 50 | $\mathrm{C}_{12} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{OCl}_{2}$ | $\mathrm{C}, \mathrm{H}, \mathrm{N}, \mathrm{Cl}$ |  | 0.5 (0.21-7.6) |
| 103 | CI | H | H | Cl | $\stackrel{C_{N}^{\prime \prime}}{C_{N}^{N-N}} \\|_{H}^{N-N}$ | 322-324 | DMF | 70 | $\mathrm{C}_{12} \mathrm{H}_{6} \mathrm{~N}_{6} \mathrm{Cl}_{2}$ | d | 0.016 (0.007-0.049) | 0.5 (0.07-22.7) |
| 104 | H | H | Cl | ${ }^{\mathrm{H}}$ | $\mathrm{CO}_{2} \mathrm{H}$ | 288-289 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 82 | $\mathrm{C}_{12} \mathrm{H}_{2} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Cl}-\mathrm{H}_{2} \mathrm{O}$ | $\mathrm{C}, \mathrm{H}, \mathrm{~N}$ | 3.74 (2.91-4.95) |  |
| 105 | Cl | H | H | $\mathrm{Me}_{2} \mathrm{CHO}$ | $\mathrm{CO}_{2} \mathrm{H}$ | 283-285 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 85 | $\mathrm{C}_{15} \mathrm{H}_{13} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{Cl}$ | $\mathrm{C}, \mathrm{H}, \mathrm{~N}, \mathrm{Cl}$ | 0.027 (0.021-0.033) | $0.34(0.07-0.40)$ |
| 106 | MeO MeO | H H | H H | $\mathrm{MeO}_{\mathrm{Me}} \mathrm{CHO}$ | $\mathrm{CO}_{2} \mathrm{CO}_{2} \mathrm{H}$ | 248-250 | $\mathrm{MeOH} / \mathrm{Et}_{2} \mathrm{O}$ $\mathrm{MeOH} / \mathrm{Et}_{2} \mathrm{O}$ | 83 | ${ }^{\mathrm{C}_{14} \mathrm{C}_{4} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{~N}_{2} \mathrm{O}_{4}}$ | $\begin{aligned} & \mathrm{C}, \mathrm{H}, \mathrm{~N} \end{aligned}$ | $1.22(0.85-1.74)$ <br> 0.091 (0.064-0.127) | inactive ${ }^{c}$ inactive ${ }^{b}$ |
| 107 | $\stackrel{\mathrm{MeO}}{\mathrm{Et}}$ | H <br> H | H H H | $\underset{\mathrm{Cl}}{\mathrm{Me}_{2} \mathrm{CHO}}$ | $\mathrm{CO}_{2} \mathrm{H}$ $\mathrm{CO}_{2} \mathrm{H}$ | 259-261 | $\mathrm{MeOH} / \mathrm{Et}_{2} \mathrm{O}$ $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 87 83 | $\xrightarrow{\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{~N}_{4} \cdot 0.5 \mathrm{~N}_{4} \mathrm{O}}$ | $\xrightarrow[\mathrm{C}]{\mathrm{C}, \mathrm{H}, \mathrm{N}, \mathrm{N}, \mathrm{Cl}}$ | $\begin{aligned} & 0.091(0.064-0.127) \\ & 0.09(0.05-0.15) \end{aligned}$ | inactive ${ }^{b}$ |
| 109 | Br | H | H | CI | $\mathrm{CO}_{2} \mathrm{H}$ | 300-303 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 91 | $\mathrm{C}_{12} \mathrm{C}_{6} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{BrCl} \cdot \mathrm{H}_{2} \mathrm{O}$ | C, $\mathrm{H}, \mathrm{N}, \mathrm{Br}, \mathrm{Cl}$ | 0.025 (0.011-0.039) | 0.08 (0.04-0.12) |
| 110 | Cl | H | H | Cl | $\mathrm{CO}_{2} \mathrm{H}$ | 304-305 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 92 | $\mathrm{C}_{12} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Cl}_{2}$ | C, $\mathrm{H}, \mathrm{N}, \mathrm{Cl}$ | 0.023 (0.019-0.026) | 0.033 (0.010-0.035) |
| 111 | MeO | H | H | Cl | $\mathrm{CO}_{2} \mathrm{Et}$ | 215-216 | $\mathrm{CHCl}_{3} / \mathrm{Et}_{2} \mathrm{O}$ | 30 | $\mathrm{C}_{15} \mathrm{H}_{13} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{Cl}$ | C,H,N, Cl |  | 8.0(4.65-18.7) |
| 112 | MeO | H | H | Cl | $\mathrm{CO}_{2} \mathrm{H}$ | 277-278 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 87 | $\mathrm{C}_{13} \mathrm{H}_{9} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{Cl}$ | C,H,N,Cl | 0.027 (0.021-0.033) | 0.34 (0.07-0.40) |
| 113 | H | Cl | H | Ph | $\mathrm{CO}_{2} \mathrm{H}$ | 307-309 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 92 | $\mathrm{C}_{18} \mathrm{H}_{11} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Cl}$ | $\mathrm{C}, \mathrm{H}, \mathrm{N}, \mathrm{Cl}$ |  | inactive ${ }^{\text {c }}$ |

Table IV. Imidazo[1,2-a]quinoxalines


| compd | $\mathrm{R}^{2}$ | $\mathrm{R}^{4}$ | $\mathrm{R}^{7}$ | $\mathrm{R}^{8}$ | $\mathrm{mp},{ }^{\circ} \mathrm{C}$ | recrystn solvent | yield, \% | formula | anal. | antiallergic activity, IgE, rat PCA:$\mathrm{ED}_{30}{ }^{a}{ }^{a} \mathrm{mg} / \mathrm{kg}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  |  |  |  | iv | po |
| 114 | $\mathrm{CO}_{2} \mathrm{H}$ | H | H | H | 274-275 | EtOH/ $\mathrm{H}_{2} \mathrm{O}$ | 95 | $\mathrm{C}_{11} \mathrm{H}_{7} \mathrm{~N}_{3} \mathrm{O}_{2}$ | C,H,N | 0.073 (0.057-0.091) | 0.13 (0.54-0.90) |
| 115 | $\mathrm{CO}_{2} \mathrm{Et}$ | H | Cl | C 1 | 297-299 | EtOH | 40 | $\mathrm{C}_{13} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{Cl}_{2}$ | C,H,N,C1 |  | inactive ${ }^{\text {b }}$ |
| 116 | $\mathrm{CO}_{2} \mathrm{H}$ | H | Cl | Cl | $>350$ | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 88 | $\mathrm{C}_{11} \mathrm{H}_{5} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{Cl}_{2}$ | C,H,N,Cl | 0.2 (0.12-0.35) | inactive ${ }^{\text {b }}$ |
| 117 | $\mathrm{CO}_{2} \mathrm{H}$ | $\mathrm{CONH}_{2}$ | H | H | 238-240 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 95 | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{~N}_{4} \mathrm{O}_{3} \cdot \mathrm{H}_{2} \mathrm{O}$ | C,H,N | 3.13 (2.48-3.45) |  |
| 118 |  | H | H | H | 280-281 | DMF | 91 | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{~N}_{8} \mathrm{O}$ | C,H,N | 0.45 (0.24-1.05) | inactive ${ }^{\text {b }}$ |
| 120 | $\mathrm{CH}_{2} \mathrm{OH}$ | H | H | H | 217-220 | $\begin{gathered} \mathrm{CHCl}_{3} / \\ \mathrm{Et}_{2} \mathrm{O} \end{gathered}$ | 53 | $\mathrm{C}_{11} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{O}$ | C,H,N | $0.156(0.066-0.453)^{c}$ | $0.061(0.021-0.089)$ |
| 121 | CHO | H | H | H | 214-216 | $\begin{array}{r} \mathrm{CHCl}_{3} / \\ \mathrm{Et}_{2} \mathrm{O} \end{array}$ | 89 | $\mathrm{C}_{11} \mathrm{H}_{7} \mathrm{~N}_{3} \mathrm{O}$ | C,H,N | $0.137(0.052-0.43)^{c}$ | 0.073 (0.031-0.15) |
| 122 |  | H | H | H | 320-322 | DMF/ $\mathrm{H}_{2} \mathrm{O}$ | 53 | $\mathrm{C}_{11} \mathrm{H}_{7} \mathrm{~N}_{7}$ | $\mathrm{C}, \mathrm{H}, \mathrm{N}^{\text {d }}$ | 0.012 (0.005-0.033) | $0.089(0.025-0.282)$ |

${ }^{a} 95 \%$ confidence limits in parentheses. ${ }^{b}$ Inactive at 1 and $10 \mathrm{mg} / \mathrm{kg} .{ }^{c}$ Tested iv as HCl salts. ${ }^{d} \mathrm{~N}$ : calcd, 41.33 ; found, 40.83 .

Scheme X. Miscellaneous Analogues ${ }^{a}$


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${ }^{a}$ Reagents: (a) (i) $\mathrm{BrCH}_{2} \mathrm{COCO}_{2} \mathrm{Et}$; DME, (ii) $\Delta$; EtOH; (b) $\mathrm{BrCH}_{2} \mathrm{COCO}_{2} \mathrm{Et}$; $\mathrm{EtOH}, \Delta$; (c) rhodanine, $\mathrm{NaOAc}, \mathrm{Ac}_{2} \mathrm{O}, \Delta$; (d) (i) $\mathrm{NaOH} ; \mathrm{H}_{2} \mathrm{O}, \mathrm{EtOH}$, (ii) HCl ; (e) $\mathrm{I}_{2}$; dioxane.
electron-withdrawing groups at position 8 (29-35) and electron-donating groups ( $\mathbf{3 6}$ and 37 ) although the iv/po ratio does vary. Also, alkyl groups in position $4(38,39)$

## Scheme XI ${ }^{a}$


${ }^{a}$ Reagents: (a) BrCN ; aqueous MeOH ; b) (i) $\mathrm{BrCH}_{2} \mathrm{COCO}_{2} \mathrm{Et}$; DME, (ii) $\Delta$; EtOH (hydrolysis: (i) $\mathrm{NaOH} ; \mathrm{H}_{2} \mathrm{O}, \mathrm{EtOH}^{2}$, (ii) HCl ).
increase the activity. Overall, the conclusion seems to be that the parent 2-carboxylic acid possesses a good level of activity which is increased in a not easily accountable manner by substituents in the 4 - and 8 -positions.
(b) Imidazo[1,2-a ]quinolines. Tables II and III list the results of testing compounds in the PCA test in comparison to DSCG. Table II lists the screening results of the benzo-unsubstituted imidazo[1,2-a]quinoline-2carboxylic acids and derivatives while Table III lists the screening results of the benzo-substituted members of the same series. In general, among all the compounds in Tables II and III, it can be seen that high levels of iv activity are present (up to 160 times that of DSCG), and furthermore, certain compounds such as 47,91 , and 110 are equiactive following iv or po dosing. In particular, concentrating on Table II, the rank order of activity for iv

Table V. Imidazo[1,2-a]quinoxalinones


| compd | $\mathrm{R}^{8}$ | $\mathrm{R}^{7}$ | $\mathrm{R}^{5}$ | $\mathrm{R}^{2}$ | $\mathrm{mp},{ }^{\circ} \mathrm{C}$ | recrystn solvent | yield, \% | formula | anal. | antiallergic activity, IgE, rat PCA: $E D_{50},{ }^{a} \mathrm{mg} / \mathrm{kg}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  |  |  |  | iv | po |
| 124 | H | H | Et | $\mathrm{CO}_{2} \mathrm{Et}$ | 216-218 | EtOH | 63 | $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{3}$ | C,H,N |  | 0.16 (0.05-0.64) |
| 125 | H | H | Pr | $\mathrm{CO}_{2} \mathrm{Et}$ | 212-214 | EtOH | 65 | $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{3}$ | C,H,N |  | 1.21 (0.74-2.02) |
| 126 | H | H | $\mathrm{CH}_{2} \mathrm{Ph}$ | $\mathrm{CO}_{2} \mathrm{Et}$ | 259-260 | DMF/EtOH | 89 | $\mathrm{C}_{20} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{3}$ | C,H,N |  | inactive $^{\text {c }}$ |
| 127 | H | H | H | $\mathrm{CO}_{2} \mathrm{H}$ | 272-275 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 95 | $\begin{gathered} \mathrm{C}_{11} \mathrm{H}_{7} \mathrm{~N}_{3} \mathrm{O}_{3} \\ 0.25 \mathrm{H}_{2} \mathrm{O} \end{gathered}$ | C,H,N | 0.16 (0.11-0.25) | 2.0 (0.97-4.61) |
| 128 | H | H | Me | $\mathrm{CO}_{2} \mathrm{H}$ | 267-268 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 95 | $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{O}_{3}$ | C,H,N | 0.031 (0.024-0.041) | 1.28 (0.67-2.51) |
| 129 | H | H | Et | $\mathrm{CO}_{2} \mathrm{H}$ | 260-262 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 79 | $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}_{3}$ | C,H,N | 0.01 (0.007-0.014) | 0.45 (0.29-0.58) |
| 130 | H | H | Pr | $\mathrm{CO}_{2} \mathrm{H}$ | 232-234 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 95 | $\begin{aligned} & \mathrm{C}_{14} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}_{3} . \\ & \mathrm{H}_{2} \mathrm{O} \end{aligned}$ | C,H,N | 0.01 (0.006-0.021) | 0.24 (0.12-0.32) |
| 131 | H | H | Bu | $\mathrm{CO}_{2} \mathrm{H}$ | 220-222 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 95 | $\begin{gathered} \mathrm{C}_{10} \mathrm{H}_{10} \mathrm{~N}_{3} \mathrm{O}_{3} . \\ \mathrm{H}_{2} \mathrm{O} \end{gathered}$ | C,H,N | 0.023 (0.058-0.12) | 0.59 (0.46-0.73) |
| 133 | Cl | Cl | H | $\mathrm{CO}_{2} \mathrm{H}$ | >330 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 95 | $\begin{gathered} \mathrm{C}_{11} \mathrm{H}_{5} \mathrm{Cl}_{2} \mathrm{~N}_{3} \mathrm{O}_{3} . \\ \mathrm{H}_{2} \mathrm{O} \end{gathered}$ | C; ${ }^{\text {b }} \mathrm{H}, \mathrm{N}$ |  | inactive ${ }^{\text {c }}$ |
| 134 | H | H | Et | $\mathrm{CH}_{2} \mathrm{OH}$ | 215-218 | EtOH | 43 | $\mathrm{C}_{13} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}_{2}$ | C,H,N | $0.071(0.05-0.11)^{\text {d }}$ | 0.062 (0.020-0.116) |
| 135 | H | H | Et | CHO | 302-306 | $\mathrm{EtOH} / \mathrm{CHCl}_{3}$ | 62 | $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}_{2}$ | C,H,N |  | 0.077 (0.028-0.149) |
| 136 | H | H | Et |  | >310 | DMF | 90 | $\begin{gathered} \mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{8} \mathrm{O}_{2} . \\ \left(\mathrm{C}_{3} \mathrm{H}_{7} \mathrm{NO}\right)^{e} \end{gathered}$ | C,H,N | 0.32 (0.25-0.41) |  |

${ }^{a} 95 \%$ confidence limits in parentheses. ${ }^{b} \mathrm{C}$ : calcd, 41.80 ; found, $42.35 .{ }^{c}$ Inactive at $10 \mathrm{mg} / \mathrm{kg} .{ }^{d}$ Tested as HCl salt. ${ }^{e}$ One mole of DMF of crystallization.

Table VI. Imidazo[1,2-a]quinazolinones and Derivatives


| compd | R | mp, ${ }^{\circ} \mathrm{C}$ | recrystn solvent | yield, \% | formula | anal. | antiallergic activity, $\operatorname{IgE}$, rat PCA: $\mathrm{ED}_{50}$, $\mathrm{mg} / \mathrm{kg}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  | iv | po |
| 137 | $\mathrm{CO}_{2} \mathrm{Et}$ | 215-217 | $\mathrm{Et}_{2} \mathrm{O}$ | 49 | $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{3}$ | C,H,N |  | inactive |
| 138 | $\mathrm{CO}_{2} \mathrm{H}$ | 279-282 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 82 | $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}_{3} \cdot 0.75 \mathrm{H}_{2} \mathrm{O}$ | C,H,N | inactive |  |

administration for compounds substituted only in the 2-position is as follows:


The activity of the primary amine 56 is surprising since this is not a group usually associated with PCA inhibitory activity. Additionally, good levels of po activity are seen for the alcohol 53 and the aldehyde 54 but, whether this is due to metabolism to the acid is unknown. It will be seen in later series that alcohols and aldehydes can show
significant levels of activity (see 120, 121, 134, and 135). For compounds with a range of substituents in the 4 position and retaining carboxyl in the 2-position, all except the $4-\mathrm{MeO}$ and $4-\mathrm{SONHMe}$ compounds, 59 and 69 , respectively, are more active than 47 , and the rank order of activity for iv administration is


By po administration only the $4-\mathrm{Br}$ compound 63 is more active than 47. Substituents in the 3 -position such as Ph (72), $\mathrm{CH}_{2} \mathrm{OH}(73)$, and $\mathrm{CN}(74)$ and those in the 1-position such as $\mathrm{Br}(76)$ and Me (78) (while retaining the 2 -carboxy group) caused a loss of activity. Therefore, as in the imidazo $[2,1-c][1,4]$ benzoxazine series, PCA inhibitory activity seems to be restricted to those tricyclics with a 2 -carboxylic acid substituent or equivalent. It should be noted that Pfizer have described the PCA inhibitory activity of a series of 4-alkoxyimidazo[1,2-a]quinoline-2-carboxylic acids. ${ }^{26}$ The only compound common to our work and

[^3]Table VII. Pyrrolo [1,2-a]quinoxalinones


|  |  |  |  |  |  |  | crystn | yield, |  |  | antiallergic activity, $\operatorname{IgE}$, rat PCA:$\mathrm{ED}_{50},{ }^{a} \mathrm{mg} / \mathrm{kg}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| compd | $\mathrm{R}^{8}$ | $\mathrm{R}^{7}$ | $\mathrm{R}^{5}$ | $\mathrm{R}^{3}$ | $\mathrm{R}^{2}$ | $\mathrm{mp},{ }^{\circ} \mathrm{C}$ | solvent | $\%$ | formula | anal. | iv | po |
| DSCG |  |  |  |  |  |  |  |  |  |  | 1.21 (1.04-1.42) | inactive |
| 142 | H | H | Me | H | $\mathrm{CO}_{2} \mathrm{Et}$ | 197-198 | EtOAc | 41 | $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{3}$ | C,H,N |  | 7.0 (3.55-11.7) |
| 143 | H | H | Bu | H | $\mathrm{CO}_{2} \mathrm{Et}$ | 146-148 | $\mathrm{Et}_{2} \mathrm{O}$ | 44 | $\mathrm{C}_{18} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{3}$ | C,H,N |  | 4.1 (1.95-11.5) |
| 145 | H | H | Me | H | $\mathrm{CO}_{2} \mathrm{H}$ | 330-333 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 95 | $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{3}$ | C,H,N | 0.089 (0.063-0.125) | 0.54 (0.45-0.63) |
| 146 | H | H | Et | H | $\mathrm{CO}_{2} \mathrm{H}$ | 314-316 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 94 | $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{3}$ | C,H,N | 0.026 (0.019-0.035) | 0.035 (0.024-0.053) |
| 147 | H | H | Pr | H | $\mathrm{CO}_{2} \mathrm{H}$ | 260-262 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 96 | $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{3}$ | C,H,N | 0.019 (0.015-0.023) | $0.032(0.023-0.047)$ |
| 148 | H | H | Bu | H | $\mathrm{CO}_{2} \mathrm{H}$ | 212-214 | EtOH/ $\mathrm{H}_{2} \mathrm{O}$ | 89 | $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{3}$ | C,H,N | 0.019 (0.014-0.024) | 0.058 (0.039-0.063) |
| 149 | H | H | Pent | H | $\mathrm{CO}_{2} \mathrm{H}$ | 184-186 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 95 | $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{3}$ | C,H,N | 0.087 (0.036-0.43) | 0.31 (0.21-0.46) |
| 150 | H | H | Ph | H | $\mathrm{CO}_{2} \mathrm{H}$ | 304-306 | EtOH | 65 | $\mathrm{C}_{18} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{3}$ | C,H,N | 1.57 (1.14-2.28) | 18.9 (12.3-33.8) |
| 151 | H | H | cyclopentyl | H | $\mathrm{CO}_{2} \mathrm{H}$ | 215-216 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 95 | $\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ | C,H,N | 0.037 (0.30-0.46) | 1.96 (1.22-2.52) |
| 152 | H | H | isopropyl | H | $\mathrm{CO}_{2} \mathrm{H}$ | 238-240 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 94 | $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ | C,H,N | 0.85 (0.49-1.41) |  |
| 153 | H | H | allyl | H | $\mathrm{CO}_{2} \mathrm{H}$ | 306-308 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 96 | $\mathrm{C}_{15} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{3}$ | C,H,N | 0.035 (0.024-0.053) | 0.28 (0.21-0.37) |
| 154 | Me | Me | Et | H | $\mathrm{CO}_{2} \mathrm{H}$ | 313-316 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 89 | $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ | C,H,N | 0.030 (0.022-0.040) |  |
| 155 | Cl | Cl | Bu | H | $\mathrm{CO}_{2} \mathrm{H}$ | 281-283 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 92 | $\mathrm{C}_{16} \mathrm{H}_{14} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{3}$ | C, $\mathrm{H}, \mathrm{Cl}, \mathrm{N}$ | 0.014 (0.009-0.024) | 0.117 (0.044-0.288) |
| 156 | Cl | H | Et | H | $\mathrm{CO}_{2} \mathrm{H}$ | 322-326 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 91 | $\mathrm{C}_{14} \mathrm{H}_{11} \mathrm{ClN}_{2} \mathrm{O}_{3}$ | C,H,N | $0.035(0.028-0.045)$ | 0.29 (0.24-0.35) |
| 157 | H | Cl | Bu | H | $\mathrm{CO}_{2} \mathrm{H}$ | 275-276 | EtOH | 93 | $\mathrm{C}_{16} \mathrm{H}_{15} \mathrm{ClN}_{2} \mathrm{O}_{3}$ | C,H,Cl,N | 0.019 (0.013-0.023) | 0.114 (0.060-0.199) |
| 158 | H | H | Et | H | $\mathrm{CH}_{2} \mathrm{OH}$ | 171-173 | EtOAc | 88 | $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{2}$ | C,H,N |  | 0.045 (0.032-0.069) |
| 159 | H | H | Et | H | CHO | 252-255 | EtoAc/Et ${ }_{2} \mathrm{O}$ | 82 | $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}$ | C,H,N |  | $2.53(1.88-2.85)$ |
| 160 | H | H | Et | H |  | 295-297 | DMF/EtOAc | 48 | $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{6} \mathrm{O}^{2}$ | C,H,N | 0.011 (0.004-0.041) | 0.44 (0.182-1.56) |
| 161 | H | H | Et | H |  | 315-318 | DMF/EtOAc | 95 | $\mathrm{C}_{15} \mathrm{H}_{13} \mathrm{~N}_{7} \mathrm{O}_{2}$ | C,H,N | 0.18 (0.13-0.26) |  |
| 163 | H | HI | Et | $\mathrm{CO}_{2} \mathrm{Me}$ | $\mathrm{CO}_{2} \mathrm{H}$ | 290-293 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 96 | $\mathrm{C}_{16} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{5}$ | C,H,N | inactive |  |

Table VIII. Pyrazolo[2,3-a]quinoxalinones


| compd | $\mathrm{R}^{5}$ | $\mathrm{R}^{2}$ | mp, ${ }^{\circ} \mathrm{C}$ | recrystn solvent | yield, \% | formula | anal. | antiallergic activity, IgE, rat PCA: $\mathrm{ED}_{50},{ }^{\circ} \mathrm{mg} / \mathrm{kg}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  |  | iv | po |
| 166 | H | $\mathrm{CO}_{2} \mathrm{H}$ | 333-334 ${ }^{\text {b }}$ | EtOH/ $\mathrm{H}_{2} \mathrm{O}$ | 93 | $\mathrm{C}_{11} \mathrm{H}_{7} \mathrm{~N}_{3} \mathrm{O}_{3}$ | C,H,N | 0.33 (0.08-1.71) |  |
| 170 | Me | $\mathrm{CO}_{2} \mathrm{H}$ | 309-312 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 54 | $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{O}_{3}$ | C,H,N | 0.19 (0.10-0.44) |  |
| 171 | Et | $\mathrm{CO}_{2} \mathrm{H}$ | 242-244 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 66 | $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}_{3}$ | C,H,N | 0.11 (0.06-0.22) |  |
| 172 | Pr | $\mathrm{CO}_{2} \mathrm{H}$ | 228-229 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 51 | $\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}_{3} \cdot \mathrm{H}_{2} \mathrm{O}$ | C,H,N | 0.023 (0.014-0.029) |  |

${ }^{a} 95 \%$ confidence limits in parentheses. ${ }^{b}$ Reference $18, \mathrm{mp}>315^{\circ} \mathrm{C}$.
Table IX. Imidazo[1,5-a]quinoxalinones, Triazolo[5,1-c][1,4]benzoxazinones, and Triazolo[1,5-a]quinoxalinones


178-181


183,185

| compd | X | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | $\mathrm{mp},{ }^{\circ} \mathrm{C}$ | recrystn solvent | yield, \% | formula | anal. | antiallergic activity, IgE, rat PCA: $E D_{50},{ }^{a} \mathrm{mg} / \mathrm{kg}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  |  |  | iv | po |
| 178 | NH | H | H | >320 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 91 | $\mathrm{C}_{11} \mathrm{H}_{7} \mathrm{~N}_{3} \mathrm{O}_{3}$ | C,H,N | 0.104 (0.073-0.151) | inactive ${ }^{\text {b }}$ |
| 179 | NH | H | Me | 302-305 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 81 | $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{O}_{3}$ | C,H,N | 0.37 (0.31-0.45) |  |
| 180 | NH | H | Et | 214-216 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 60 | $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}_{3}$ | C,H,N | inactive ${ }^{\text {c }}$ |  |
| 181 | NH | H | $\mathrm{CH}_{2} \mathrm{Ph}$ | 181-183 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 47 | $\mathrm{C}_{18} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}_{3}$ | C,H,N |  | inactive ${ }^{\text {c }}$ |
| 183 | 0 | Me |  | 205-207 | EtOAc | 54 | $\mathrm{C}_{11} \mathrm{H}_{7} \mathrm{~N}_{3} \mathrm{O}_{4}$ | C,H,N |  | inactive ${ }^{\text {d }}$ |
| 185 | NH | H |  | >320 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 76 | $\mathrm{C}_{10} \mathrm{H}_{6} \mathrm{~N}_{4} \mathrm{O}_{3}$ | C,H,N | 0.084 (0.058-0.12) | inactive ${ }^{e}$ |

${ }^{a} 95 \%$ confidence limits in parentheses. ${ }^{b}$ Inactive at $20 \mathrm{mg} / \mathrm{kg}$. ${ }^{c}$ Inactive at 1 and $10 \mathrm{mg} / \mathrm{kg}$. ${ }^{d} 50 \%$ inhibition at $10 \mathrm{mg} / \mathrm{kg}$. ${ }^{e} 45 \%$ inhibition at $20 \mathrm{mg} / \mathrm{kg}$.

Table X. Related Ring Systems


186


188


190


191

195.197


199,201,203,205

|  |  |  |  |  |  |  |  |  |  | antiallergic activity, IgE , rat PCA: $\mathrm{ED}_{50}$, ${ }^{a}$ $\mathrm{mg} / \mathrm{kg}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| compd | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | $\mathrm{R}^{3}$ | $\mathrm{R}^{4}$ | mp, ${ }^{\circ} \mathrm{C}$ | solvent | \% | formula | anal. | iv | po |
| 186 |  |  |  |  | 175-178 ${ }^{\text {b }}$ | $\mathrm{EtOH} / \mathrm{Et}_{2} \mathrm{O}$ | 31 | $\mathrm{C}_{10} \mathrm{H}_{11} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{Br}$ |  | inactive ${ }^{c}$ |  |
| 188 |  |  |  |  | 242-247 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 82 | $\mathrm{C}_{13} \mathrm{H}_{9} \mathrm{NO}_{2}$ | C,H,N | inactive ${ }^{c}$ |  |
| 190 |  |  |  |  | $>370^{\text {d }}$ | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 80 | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{O}_{2}$ | C,H,N | inactive ${ }^{\text {c }}$ |  |
| 191 |  |  |  |  | 281-284 ${ }^{\text {e }}$ | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 66 | $\mathrm{C}_{13} \mathrm{H}_{8} \mathrm{O}_{2} \mathrm{~S}$ | C,H,S | inactive ${ }^{\text {c }}$ |  |
| 195 | Ph | $\mathrm{CO}_{2} \mathrm{H}$ |  |  | 181-183 | $\mathrm{MeOH} / \mathrm{Et}_{2} \mathrm{O}$ | 96 | $\mathrm{C}_{16} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ | C,H,N,S | 1.92 (1.07-3.93) |  |
| 197 | Me | $\mathrm{CO}_{2} \mathrm{H}$ |  |  | 196-200 | DMF/EtOAc | 85 | $\mathrm{C}_{11} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ | C,H,N,S | 2.59 (2.20-3.00) |  |
| 199 | H | H | H | $\mathrm{CO}_{2} \mathrm{H}$ | 254-255 ${ }^{\text {f }}$ | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 91 | $\mathrm{C}_{10} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}$ | C,H,N,S | 0.33 (0.19-0.59) |  |
| 201 | H | H | MeO | $\mathrm{CO}_{2} \mathrm{H}$ | 270-273 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 86 | $\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S} \cdot 1.25 \mathrm{H}_{2} \mathrm{O}$ | C,H,N,S | 0.62 (0.39-0.96) |  |
| 203 | H | Me | Me | $\mathrm{CO}_{2} \mathrm{H}$ | 307-308 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 82 | $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S} \cdot 0.33 \mathrm{H}_{2} \mathrm{O}$ | C,H,N,S | 0.30 (0.22-0.38) |  |
| 205 | MeO | H | H | $\mathrm{CO}_{2} \mathrm{H}$ | 288-290 | $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ | 85 | $\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$ | C,H,N,S | 0.16 (0.072-0.41) |  |

${ }^{a} 95 \%$ confidence limits in parentheses. ${ }^{b}$ Reference $22, \mathrm{mp} 174.5-175.5^{\circ} \mathrm{C} .{ }^{c}$ Inactive at 1 and $10 \mathrm{mg} / \mathrm{kg}$. ${ }^{d}$ Reference 23. ${ }^{e}$ Reference 24 , mp 277-278 ${ }^{\circ} \mathrm{C}$. ${ }^{\boldsymbol{t}}$ Reference $25, \mathrm{mp} 263-265^{\circ} \mathrm{C}$.
theirs is 59 for which we find very similar levels of activity, i.e., $78 \%$ inhibition at $0.3 \mathrm{mg} / \mathrm{kg}$ iv and $90 \%$ inhibition at $3.0 \mathrm{mg} / \mathrm{kg}$ po against $\mathrm{ED}_{50^{\prime} \text { s }}$ of 0.57 and 3.74 , respectively (Table II). While some 4 -chloro compounds were described
in the patent, their activities were not given, and so our finding that compounds with $4-\mathrm{Cl}(62), 4-\mathrm{Br}$ (63), and $4-\mathrm{SMe}$ (64) substituents are more active than the $4-\mathrm{MeO}-$ substituted compound is not corroborated.

Table III lists the activities of compounds with substituents in the benzo ring. Substituents in the 7 - and especially 8-positions enhance activities, the rank order for the 8 -position for iv administration being:

| $\mathrm{SO}_{2} \mathrm{Me}>\mathrm{MeO}>\mathrm{Cl}>\mathrm{SMe}=\mathrm{SOMe}=\mathrm{SONHMe}=\mathrm{CF}_{3}>\mathrm{Et}>\mathrm{H}$ |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{8 7}$ | $\mathbf{8 1}$ | $\mathbf{8 0}$ | $\mathbf{8 3}$ | 85 | 88 | 90 | 82 | 47 |

Disubstitution, especially in the 5,8 -positions, also increased iv and po activities and compounds 109 and 110 have the best overall combined iv and po activities of this series.
(c) Imidazo[1,2-a ]quinoxalines. The best results (Table IV) in this limited series were again found with acid and tetrazole groups in the 2 -position. Substitution in the benzene ring ( 115,116 ) or in the 4 -position (117) reduced activity.
(d) Imidazo[1,2-a ]quinoxalinones. The esters were too insoluble to be tested by iv administration (Table V), but in the acid series the iv activity increases with N -alkyl chain length to a maximum at $n$-propyl (130). However, the oral activity was not as good as in some of the other series.
(e) Imidazo1,2-a ]quinazolinones. The "reverse amide" analogues ( 137,138 ) were inactive (Table VI), indicating the limits to change allowed in the 4,5 -positions.
(f) Pyrrolo[1,2-a ]quinoxalinones. Table VII shows that all compounds of this series (other than the acid ester 163) show significant activity in the PCA test following either, and in some cases both, iv or po dosing.
Indeed many compounds show activity of the order of 100 times that of DSCG. In the 2 -carboxylic acid series, the activity peaks after iv administration with the $5-\mathrm{Pr}$ and $5-\mathrm{Bu}$ substituents $(147,148)$ and after po dosing with the $5-\mathrm{Et}$ and $5-\mathrm{Pr}$ groups (146, 147). Maximal activity is restricted to straight-chain alkyl or alkenyl substituents on the 5 -nitrogen atom, and the presence of branched chains as in 151 and 152 or a phenyl ring as in 150 in this position causes a dramatic loss of activity, perhaps indicating a subtle balance of steric and lipophilic effects. Surprisingly, the esters 142 and 143 show little po activity compared to the corresponding acids 145 and 148. The introduction of substituents in the aromatic ring, e.g., 154-157, has little effect on iv activities but generally causes a slight drop in po activities.
(g) Pyrazolo[2,3-a ]quinoxalin-4(5H)-ones. As shown in Table VIII, the compounds were inactive orally and were less active following iv administration than their imida-zo[1,2-a]quinoxalinone counterparts (compare 170 and 145; 171 and 146; 172 and 147).
(h) Imidazo[1,5-a ]quinoxalinones, Triazolo[1,5-a ]quinoxalinones, and Triazolo[1,5-a ]benzoxazinones. Table IX shows the activities found in these compound in which the acid or ester group is now in the 3-position. While 178, 179, and 185 showed reasonable intravenous activity, none of the compounds showed any oral activity.
(i) Miscellaneous Ring Systems. The results from the screening of compounds from related ring systems are shown in Table X. No activity was found for compounds $186,188,190$, and 191 . The two cyclic sulfoximide acids 195 and 196 showed moderate levels of iv activity and slightly better were the imidazobenzothiazoles 199, 201, 203, and 205.

## Conclusions

In summary, very good levels of activity in the IgEmediated PCA test following either iv or po doses (or in some cases both) have been found in 2-carboxylic acids derived from $4 H$-imidazo [2,1-c][1,4]benzoxazines (Table I), imidazo[1,2-a]quinolines (Table II and III), imidazo-
[1,2-a]quinoxalines (Table IV), imidazo[1,2-a]quinoxalinones (Table V), pyrrolo[1,2-a]quinoxalinones (Table VII), pyrazolo[2,3-a]quinoxalinones (Table VIII), imidazo[1,5-a]quinoxalinones (Table IX), triazolo[1,5-a]quinoxalinones (Table IX), imidazo[1,2-c][1,2,4]benzothiadiazines (Table X), and imidazo $[1,2-a]$ benzothiazoles (Table X). In addition, it is possible to replace the carboxylic acid group by alcohol, aldehyde, tetrazoyl, tetrazoylamide, and aminomethyl. Within an active series, a certain number of benzo substituents are allowed without destroying the activity, but the effects are not easily explicable in terms of lipophilicity or electronic effects. Of these series, the two belonging to the imidazo $[1,2-a]$ quinoline ${ }^{27}$ and imidazo $\left[1,2-a\right.$ ]quinoxalines ${ }^{28}$ show the best activities, and the acid 114 has been chosen for further development both on the basis of the results described here and the lack of cross tachyphylaxis with DSCG. ${ }^{29}$

## Experimental Section

Melting points were determined with an Electrothermal open-ended capillary melting point apparatus and are uncorrected. IR spectra were recorded on a Unicam SP1000 infrared spectrophotometer as KBr disks. ${ }^{1} \mathrm{H}$ NMR spectra were recorded in either $\mathrm{CDCl}_{3}$ or $\mathrm{Me}_{2} \mathrm{SO}-d_{6}$ with a Perkin-Elmer R12A spectrometer ( 60 MHz ) with $\mathrm{Me}_{4} \mathrm{Si}$ as internal standard. All OH and NH peaks were exchanged on $\mathrm{D}_{2} \mathrm{O}$ shake. IR and NMR spectra were run on all compounds and were fully in accord with the assigned structures. Elemental analyses (determined by CHN Analysis Ltd., Leicester, England) were carried out on all new compounds and results were within $0.4 \%$ of the expected values, except where noted. All organic extracts were dried over magnesium sulfate and solid products were dried under vacuum over $\mathrm{P}_{2} \mathrm{O}_{5}$.
(a) Imidazo[2,1-c $][1,4]$ benzoxazines. ${ }^{9} \boldsymbol{N}$-( $1 \boldsymbol{H}$-Tetrazol-$5-\mathrm{yl})-4 \mathrm{H}$-imidazo $[2,1-\mathrm{c}][1,4]$ benzoxazine-2-acetamide (26). 1, $1^{\prime}$-Carbonyldiimidazole ( $0.90 \mathrm{~g}, 55 \mathrm{mmol}$ ) was added to a stirred solution of the acetic acid $17^{9}(1.15 \mathrm{~g}, 50 \mathrm{mmol})$ in DMF ( 20 mL ) and the mixture was stirred at room temperature for 5 h . Anhydrous 5 -aminotetrazole ( $0.5 \mathrm{~g}, 59 \mathrm{mmol}$ ) was added and stirring was continued overnight. The mixture was then concentrated by evaporation in vacuo (air-bleed) and triturated with $\mathrm{CHCl}_{3}$ to give $26(750 \mathrm{mg}, 44 \%)$ as pale pink crystals, $\mathrm{mp} 282-284^{\circ} \mathrm{C}$ (from DMF-CHCl ${ }_{3}$ ). Anal. $\left(\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{~N}_{7} \mathrm{O}_{2}\right.$ ) C, $\mathrm{H}, \mathrm{N}$.

2-[(1H-5-Aminotetrazol-1-yl)carbonyl]-4H-imidazo[2,1c ] [1,4]benzoxazine (24) and $\boldsymbol{N}$-( $1 \boldsymbol{H}$-Tetrazol-5-yl)-4H-imidazo[2,1-c][1,4]benzoxazine-2-carboxamide (25). Reaction of the carboxylic acid $3^{9}$ with 5 -aminotetrazole as in the preparation of 26 gave the carboxamide $\mathbf{2 5}, \mathrm{mp} 296-297^{\circ} \mathrm{C}$; IR 740 , $1257,1516,1537,1570,1609,1700,3140$, and $2500-3250 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $\mathrm{Me}_{2} \mathrm{SO}-d_{6}$ ) $\delta 5.33\left(2 \mathrm{H}, \mathrm{s}, 4-\mathrm{H}\right.$ ), 6.8-7.2 ( $3 \mathrm{H}, \mathrm{m}, \mathrm{Ar}^{\prime} \mathrm{H}$ ), 7.6-7.95 ( $1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}$ ), $8.72(1 \mathrm{H}, \mathrm{s}, 1-\mathrm{H})$. Anal. ( $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{7} \mathrm{O}_{2}$ ) $\mathrm{C}, \mathrm{H}, \mathrm{N}$.

Reaction of the acid chloride of 3 with 5 -aminotetrazole under azeotroping conditions in DMF-benzene yielded a mixture of 25 and a similar compound separated by fractional crystallization from MeOH and tentatively assigned structure 24: mp 300-301 ${ }^{\circ} \mathrm{C}$; IR 750, 1242, 1506, 1560, 1603, 1690, 3130, 3280, and 3395 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{Me}_{2} \mathrm{SO}-d_{6}\right) \delta 5.30(2 \mathrm{H}, \mathrm{s}, 4-\mathrm{H}), 6.95-7.2(3 \mathrm{H}, \mathrm{m}$, $\operatorname{ArH}), 7.5-7.8(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}), 8.60(1 \mathrm{H}, \mathrm{s}, 1-\mathrm{H})$. Anal. $\left(\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{7} \mathrm{O}_{2}\right)$ C, $\mathrm{H}, \mathrm{N}$.

Methyl 1-(Hydroxymethyl)-4H-imidazo[2,1-c][1,4]benz-oxazine-2-carboxylate (27). Sodium borohydride $(1.0 \mathrm{~g}, 26$ mmol ) was added slowly over a 2 -h period to a solution of the acid chloride of $19^{9}(0.85 \mathrm{~g}, 2.9 \mathrm{mmol})$ in dimethoxyethane ( 20 mL ) at room temperature. Stirring of the reaction mixture was continued for a further 4 h . The mixture was then poured into water ( 100 mL )- $\mathrm{CHCl}_{3}\left(100 \mathrm{~mL}\right.$ ). The $\mathrm{CHCl}_{3}$ layer was removed and the aqueous layer was further extracted with $\mathrm{CHCl}_{3}(2 \times 50$ mL ). The combined $\mathrm{CHCl}_{3}$ extract was washed once with water, dried, and evaporated. Trituration with $\mathrm{Et}_{2} \mathrm{O}$ gave 27 as colorless

[^4]needles ( $0.35 \mathrm{~g}, 46 \%$ ), mp 209-210 ${ }^{\circ} \mathrm{C}$ (from $\mathrm{Et}_{2} \mathrm{O}$ ). Anal. $\left(\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{4}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

1-(Hydroxymethyl)-4H-imidazo[2,1-c ][1,4]benzoxazine-2-carboxylic Acid (28). Ester 27 was hydrolyzed as for the preparation of 47 to give acid $28(78 \%)$, mp $232-233^{\circ} \mathrm{C}$ (from EtOH- $\mathrm{H}_{2} \mathrm{O}$ ). Anal. ( $\left.\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{4} \cdot \mathrm{O} \cdot 5 \mathrm{H}_{2} \mathrm{O}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.
The following compounds were prepared in the same wey as $3^{9}$ in yields of $75-95 \%$.

8-Chloro-4 $\boldsymbol{H}$-imidazo [2,1-c] [1,4]benzoxazine-2-carboxylic acid (34), mp $276-278{ }^{\circ} \mathrm{C}$ (from EtOH). Anal. $\left(\mathrm{C}_{11} \mathrm{H}_{7} \mathrm{ClN}_{2} \mathrm{O}_{3}\right.$ ) $\mathrm{C}, \mathrm{H}, \mathrm{Cl}, \mathrm{N}$.

Methyl 8-chloro-4H-imidazo[2,1-c][1,4]benzoxazine-2carboxylate (35), mp 239-240 ${ }^{\circ} \mathrm{C}$ (from $\mathrm{EtOH}-\mathrm{Et}_{2} \mathrm{O}$ ). Anal. $\left(\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{ClN}_{2} \mathrm{O}_{3}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

Methyl 8 -methyl- $4 \boldsymbol{H}$-imidazo [2,1-c][1,4]benzoxazine-2carboxylate (36), mp $195-197^{\circ} \mathrm{C}$ (from $\mathrm{Et}_{2} \mathrm{O}$ ). Anal. ( $\mathrm{C}_{13}{ }^{-}$ $\mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{3}$ ) C, H, N.

8-Methyl-4 $\boldsymbol{H}$-imidazo[2,1-c ][1,4]benzoxazine-2-carboxylic acid (37), mp 221-223 ${ }^{\circ} \mathrm{C}$ (from EtOH- $\mathrm{H}_{2} \mathrm{O}$ ). Anal. ( $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{3}$ ) C, H,N.

4-Methyl-4 $\boldsymbol{H}$-imidazo[2,1-c $][1,4]$ benzoxazine-2-carboxylic acid (38), mp 210-211 ${ }^{\circ} \mathrm{C}$ (from EtOH- $\mathrm{H}_{2} \mathrm{O}$ ). Anal. ( $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{3}$ ) $\mathrm{C}, \mathrm{H}, \mathrm{N}$.

4-Ethyl-4 $\boldsymbol{H}$-imidazo[2,1-c ][1,4]benzoxazine-2-carboxylic acid (39), mp $172-174^{\circ} \mathrm{C}$ (from $\mathrm{EtOH}-\mathrm{H}_{2} \mathrm{O}$ ). Anal. $\left(\mathrm{C}_{13} \mathrm{H}_{12}{ }^{-}\right.$ $\mathrm{N}_{2} \mathrm{O}_{3} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ ) $\mathrm{C}, \mathrm{H}, \mathrm{N}$.
(b) Imidazo[1,2-a ]quinolines. The following substituted 2-aminoqumolines were prepared according to the routes described in the literature; 4-hydroxy, ${ }^{30} 4$-methoxy, ${ }^{26,31} 4$-isopropoxy, ${ }^{26,31}$ 4 -chloro, ${ }^{32} 4$-bromo, ${ }^{33} 6$-chloro-4-phenyl, ${ }^{34} 3$-phenyl, ${ }^{35} 3$-ethoxycarbonyl, ${ }^{36} 7$-chloro, ${ }^{37} 7$-methoxy, ${ }^{37}$ 4-chloro-7-ethyl, ${ }^{26}$ 7methylthio, ${ }^{26} 6$-chloro, ${ }^{38} 6$-nitro, ${ }^{39} 7$-ethyl-4-hydroxy, ${ }^{26} 7$-bromo-4-hydroxy, ${ }^{26}$ 7-chloro-4-hydroxy, ${ }^{32}$ 4-hydroxy-7-methoxy, ${ }^{32}$ 7-chloro-4-isopropoxy, ${ }^{26,30} 4,7$-dimethoxy, ${ }^{26} 4$-isopropoxy- 7 -methoxy, ${ }^{26} 7$-bromo-4-chloro, ${ }^{26} 4,7$-dichloro, ${ }^{32}$ and 4-chloro-7-methoxy. ${ }^{26}$
2-Amino-4-(methylthio) quinoline. A solution of sodium (2.0 g, 87 mg -atom) dissolved in EtOH ( 500 mL ) was cooled in an ice bath and treated with a slow stream of MeSH for 0.5 h . 2-Amino-4-chloroquinoline ( $5.0 \mathrm{~g}, 28 \mathrm{mmol}$ ) was added and the mixture was refluxed for 36 h (passing the effluent gases through acidic $\mathrm{KMnO}_{4}$ solution). The solution was then distilled under a stream of nitrogen, reducmg the volume to 50 mL . The mixture was treated with ice/water ( 700 mL ) and allowed to stand 1 h to crystallize buff crystals of the (methylthio)quinoline ( 5.0 g , $94 \%$ ), mp 173-175 ${ }^{\circ} \mathrm{C}$ (from $\mathrm{CHCl}_{3}-\mathrm{Et}_{2} \mathrm{O}$ ). Anal. $\left(\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{~S}\right.$ ) C, H, N, S.

2-Amino-4-(methylsulfinyl)quinoline. A solution of 2-amino-4-(methylthio)quinoline ( $12.0 \mathrm{~g}, 63 \mathrm{mmol}$ ) in EtOH ( 1500 mL ) was stirred vigorously while being heated under reflux, and $\mathrm{NaIO}_{4}(18 \mathrm{~g}, 84 \mathrm{mmol})$ in water $(250 \mathrm{~mL})$ was added in portions over 24 h . The solution was then cooled and evaporated to a small volume. Water ( 600 mL ) was added and the solution stood for 1 h to crystallize the (methylsulfinyl)quinoline ( $8.6 \mathrm{~g}, 66 \%$ ), as light brown crystals, $\mathrm{mp} 238-240^{\circ} \mathrm{C}$ (from $\mathrm{CHCl}_{3}-\mathrm{MeOH}-\mathrm{Et}_{2} \mathrm{O}$ ). Anal. $\left(\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{OS}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}, \mathrm{S}$.

2-Amino-3-(hydroxymethyl)quinoline. Ethyl 2-amino-quimoline-3-carboxylate ${ }^{36}(8 \mathrm{~g}, 30 \mathrm{mmol})$ was dissolved in dry THF
(30) Grout, R. J.; Hyman, B. M.; Partridge, M. W. J. Chem. Soc., Perkin Trans. 1 1973, 1314.
(31) Hardman, R.; Partridge, M. W. J. Chem. Soc. 1955, 510.
(32) Hardman, R.; Partridge, M. W. J. Chem. Soc. 1958, 614.
(33) Den Hertog, H. J.; Buurman, D. J. Recl. Trav. Chim. Pays-Bas 1972, $91,841$.
(34) Kwan, S.; Tanaka, S.; Isagawa, K. Yuki Gosei Kagaku Kyokaishi 1973, 31(4), 328.
(35) Bauer, K. H. Chem. Ber. 1938, 71B, 2226. Dietrick, D. K.; Hawes, E. M.; Watteyne, G. L. Can. J. Pharm. Sci. 1971, 6(2), Haw
50.
(36) Higashino, T.; Ito, H.; Hayashi, E. Chem. Pharm. Bull. 1972, 20(7), 1544.
(37) Cotrel, C.; Jeanmart, C.; Messer M. N. U.K. Patent 1367589 , 8 Jan. 1973.
(38) Homer, J. K.; Henry, D. W. J. Med. Chem. 1968, 11 (5), 846.
(39) Simpson, J. C. E.; Wright, P. H. J. Chem. Soc. 1948, 1707.
$(200 \mathrm{~mL})$ and added dropwise to $\mathrm{LiAlH}_{4}(4 \mathrm{~g}, 105 \mathrm{mmol})$ in dry THF ( 200 mL ) heated under reflux. Heating was continued for 1 h , and then the mixture was cooled to room temperature and water ( 200 mL ) was added dropwise. The solution was extracted with EtOAc $(2 \times 300 \mathrm{~mL})$ and the combined EtOAc extract was washed once with water, dried, and evaporated to give the (hydroxymethyl) quinoline ( $4.3 \mathrm{~g}, 68 \%$ ), as off-white crystals, mp $197-199^{\circ} \mathrm{C}$ (from $\mathrm{MeOH}-\mathrm{Et}_{2} \mathrm{O}$ ). Anal. $\left(\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

2-Amino-7-(trifluoromethyl)quinoline. 2-Chloro-7-(trifluoromethyl)quinoline ( $6.5 \mathrm{~g}, 28 \mathrm{mmol}$ ) [prepared ${ }^{40}$ from 3 (trifluoromethyl)aniline] was suspended in aqueous ammonia ( $d$ $0.88,75 \mathrm{~mL}$ ) in a pressure vessel containing a small quantity of $\mathrm{Cu}_{2} \mathrm{Cl}_{2}(250 \mathrm{mg})$. The vessel was shaken and heated to $150^{\circ} \mathrm{C}$, with the pressure reaching 18 atm . After 6 h , the vessel was allowed to cool and the ammonia solution was reduced in volume and cooled. The crude precipitated product was filtered, dissolved in $\mathrm{CHCl}_{3}(200 \mathrm{~mL})$, and extracted with $2 \mathrm{~N} \mathrm{HCl}(2 \times 100 \mathrm{~mL})$. The acid extract was made basic with $\mathrm{Na}_{2} \mathrm{CO}_{3}$ and extracted with $\mathrm{CHCl}_{3}(2 \times 100 \mathrm{~mL})$. The combined $\mathrm{CHCl}_{3}$ extract was dried and evaporated and the residue was triturated with $\mathrm{CHCl}_{3}-\mathrm{Et}_{2} \mathrm{O}$ to give the (trifluoromethyl)quinoline ( $2.56 \mathrm{~g}, 41 \%$ ), mp 174-177 ${ }^{\circ} \mathrm{C}$ (from $\mathrm{CHCl}_{3}-\mathrm{Et}_{2} \mathrm{O}$ ). Anal. $\left(\mathrm{C}_{10} \mathrm{H}_{7} \mathrm{~F}_{3} \mathrm{~N}_{2}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

2-Amino-5-chloroquinoline. 2-Chloro-6-nitroanline ${ }^{41}(8.6 \mathrm{~g}$, 50 mmol ) suspended in concentrated $\mathrm{HCl}(10.8 \mathrm{~mL})$ and AcOH $(30 \mathrm{~mL})$ was stirred at -2 to $2^{\circ} \mathrm{C}$ during the addition of $\mathrm{NaNO}_{2}$ $(3.5 \mathrm{~g}, 50.8 \mathrm{mmol})$ in water $(10 \mathrm{~mL})$ over 2 h . A further quantity of $\mathrm{NaNO}_{2}(200 \mathrm{mg}, 2.9 \mathrm{mmol})$ in water $(1 \mathrm{~mL})$ was added and the mixture was stirred for a further 1 h and then treated with urea (to destroy any remaining nitrous acid). After filtration (cold) through Celite, the filtrate was slowly added to acrylonitrile (13.9 $\mathrm{g}, 262 \mathrm{mmol}) \mathrm{m} \mathrm{Me} \mathrm{M}_{2} \mathrm{CO}(100 \mathrm{~mL})$ at $0-4^{\circ} \mathrm{C}$. A suspension of CuCl $(0.5 \mathrm{~g}, 5.05 \mathrm{mmol})$ and $\mathrm{LiCl}(0.3 \mathrm{~g}, 7.1 \mathrm{mmol})$ in $\mathrm{Me}_{2} \mathrm{CO}(30 \mathrm{~mL})$ was then added, while the temperature was maintained at $0^{\circ} \mathrm{C}$. After the mixture was allowed to warm up to room temperature, the $\mathrm{Me}_{2} \mathrm{CO}$ was removed under reduced pressure and the residue was dissolved in $\mathrm{Et}_{2} \mathrm{O}(300 \mathrm{~mL})$. The $\mathrm{Et}_{2} \mathrm{O}$ solution was washed with water, dilute $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution, and water and then dried and evaporated. Chromatography of the red oil so obtained $\left(\mathrm{SiO}_{2}\right.$; $10 \%$ EtOAc-60-80 ${ }^{\circ} \mathrm{C}$ petroleum ether) gave 2-chloro-6-nitrodihydrocinnamonitrile as a pale yellow oil ( $1.7 \mathrm{~g}, 14 \%$ ).

The yellow oil ( $1.7 \mathrm{~g}, 6.95 \mathrm{mmol}$ ) was heated under reflux for 5 h in benzene $(50 \mathrm{~mL})$ in the presence of iron powder $(8 \mathrm{~g}, 143$ mmol ) while concentrated $\mathrm{HCl}(0.4 \mathrm{~mL})$ and water ( 1 mL ) were added slowly. The solution was then filtered hot and the residue was made basic with $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution before being extracted with benzene ( $2 \times 100 \mathrm{~mL}$ ). The benzene solutions were combined and washed with $2 \mathrm{~N} \mathrm{HCl}(4 \times 50 \mathrm{~mL})$. The aqueous acid extract was washed with EtOAc ( 100 mL ), basified with $\mathrm{Na}_{2} \mathrm{CO}_{3}$, and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 100 \mathrm{~mL})$. The $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution was washed with water, dried, and evaporated to give a pale brown solid ( 440 mg ) after trituration with $\mathrm{Et}_{2} \mathrm{O}-40-60^{\circ} \mathrm{C}$ petroleum ether. The solid was chromatographed $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}\right)$ to give 2-amino-5-chloroquinoline ( $180 \mathrm{mg}, 14.5 \%$ ) as off-white needles, $\mathrm{mp} 176-178{ }^{\circ} \mathrm{C}$ (from EtOAc-Et $\mathrm{E}_{2} \mathrm{O}$ ). Anal. $\left(\mathrm{C}_{9} \mathrm{H}_{7} \mathrm{~N}_{2} \mathrm{Cl}\right) \mathrm{H}, \mathrm{N}$, Cl ; C : calcd, 60.6 ; found, 59.9

1-(Ethoxalylmethyl)quinolinium Bromide (43). Quinoline $(5.0 \mathrm{~g}, 39 \mathrm{mmol})$ was dissolved in a mixture of dimethoxyethane (DME) $(25 \mathrm{~mL})-\mathrm{Et}_{2} \mathrm{O}(25 \mathrm{~mL})$ and ethyl bromopyruvate $(8.0 \mathrm{~g}$, 40 mmol ) was added. The solution was kept at room temperature for 16 h to crystallize the quaternary salt $43(4.9 \mathrm{~g}, 38 \%), \mathrm{mp}$ $128-130^{\circ} \mathrm{C}$ (from $\mathrm{EtOH}-\mathrm{Et}_{2} \mathrm{O}$ ); IR 783, 1147, 1537, 1654, and $2900-3250 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{Me}_{2} \mathrm{SO}-d_{6}\right) \delta 0.93$ and $1.18(2 \times 3 \mathrm{H}$, $2 \times \mathrm{t}, J=7 \mathrm{~Hz}$, ester $\mathrm{CH}_{3}$ and ethanol $\left.\mathrm{CH}_{3}\right), 3.39$ and $4.16(2 \times$ $2 \mathrm{H}, 2 \times \mathrm{q}, J=7 \mathrm{~Hz}$, ester $\mathrm{CH}_{2}$ and ethanol $\left.\mathrm{CH}_{2}\right), 5.49(2 \mathrm{H}$, br $\left.\mathrm{s}, \mathrm{NCH}_{2}\right), 7.9-8.8(5 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$, and $9.45(2 \mathrm{H}, \mathrm{brd}, 2-\mathrm{H})$. Anal. $\left(\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{BrNO}_{3} \cdot \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}\right) \mathrm{C}, \mathrm{H}, \mathrm{Br}, \mathrm{N}$.

Ethyl 4,5-Dihydroimidazo[1,2-a ]quinoline-2-carboxylate (44) and Ethyl Imidazo[1,2-a ]quinoline-2-carboxylate (45). The quaternary salt $43(3.5 \mathrm{~g}, 10.8 \mathrm{mmol})$ was dissolved in AcOH ( 30 mL ) and $\mathrm{NH}_{4} \mathrm{OAc}(6 \mathrm{~g}, 78 \mathrm{mmol})$ was added. After the mixture had been heated under reflux for 4 h , the solution was

[^5]poured into water ( 200 mL ), basified to $\mathrm{pH} 9-10\left(\mathrm{Na}_{2} \mathrm{CO}_{3}\right)$, and extracted with $\mathrm{CHCl}_{3}(3 \times 100 \mathrm{~mL})$. The combined $\mathrm{CHCl}_{3}$ extract was washed with water ( 100 mL ), dried, and evaporated. The oil so obtained was chromatographed ( $\left.\mathrm{SiO}_{2} ; \mathrm{EtOA} c\right)$ to give 44 , as the lower $R_{f}$ product ( $0.65 \mathrm{~g}, 25 \%$ ): mp $134-135^{\circ} \mathrm{C}$ (from
 $\left(\mathrm{CDCl}_{3}\right) \delta 1.40\left(3 \mathrm{H}, \mathrm{t}\right.$, ester $\left.\mathrm{CH}_{3}\right), 3.0-3.25(4 \mathrm{H}, \mathrm{br} \mathrm{m}, 4-\mathrm{H}$ and $5-\mathrm{H}) 4.40\left(2 \mathrm{H}, \mathrm{q}\right.$, ester $\left.\mathrm{CH}_{2}\right), 7.25-7.50(4 \mathrm{H}, \mathrm{br} \mathrm{m}, \mathrm{Ar} \mathrm{H})$, and $8.05(1 \mathrm{H}, \mathrm{s}, 1-\mathrm{H})$. Anal. $\left(\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{2}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$. The higher $R$ product was 45 ( $0.92 \mathrm{~g}, 35 \%$ ): mp 175-177 ${ }^{\circ} \mathrm{C}$ (from EtOAc-Et $\mathrm{E}_{2} \mathrm{O}$ ); IR 745, 1249, 1538, 1571, 1702, and $3145 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $\delta 1.45\left(3 \mathrm{H}, \mathrm{t}\right.$, ester $\left.\mathrm{CH}_{3}\right), 4.49\left(2 \mathrm{H}, \mathrm{q}\right.$, ester $\left.\mathrm{CH}_{2}\right), 7.45-8.10$ (6 $\mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ), and $8.69(1 \mathrm{H}, \mathrm{s}, 1-\mathrm{H})$. Anal. $\left(\mathrm{C}_{14} \mathrm{H}_{17} \mathrm{~N}_{2} \mathrm{O}_{2}\right) \mathrm{C}, \mathrm{H}$ N.

Alternatively, 2-aminoquinoline ( $30 \mathrm{~g}, 0.208 \mathrm{~mol}$ ) was dissolved in DME ( 300 mL ) with warming and a solution of ethyl bromo pyruvate ( $50 \mathrm{~g}, 0.256 \mathrm{~mol}$ ) in DME ( 50 mL ) was added. The solution was cooled in an ice bath for 1 h and the yellow precipitate was filtered and washed well with $\mathrm{Et}_{2} \mathrm{O}$. A suspension of the salt $(45 \mathrm{~g})$ in $\mathrm{EtOH}(200 \mathrm{~mL})$ was heated under reflux for 2 h , reduced in volume on a rotary evaporator, and triturated with $\mathrm{Et}_{2} \mathrm{O}$. The resulting HBr salt of 45 was filtered and dissolved in water (250 mL ). The solution was basified with dilute $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution and extracted with $\mathrm{CHCl}_{3}(3 \times 100 \mathrm{~mL})$. The combined $\mathrm{CHCl}_{3}$ extract was decolorized (charcoal) and evaporated. Trituration with $\mathrm{Et}_{2} \mathrm{O}$ gave 45 ( $26.8 \mathrm{~g}, 55 \%$ overall from 2 -aminoquinoline) identical in all respects with that prepared from 43. In this way substituted 2 -aminoquinolines were used to prepare the imidazo[1,2-a]-quinoline-2-carboxylates, including $61,79,100$, and 111.

Imidazo[1,2-a ]quinoline-2-carboxylic Acid (47). Ester 45 ( $2.0 \mathrm{~g}, 8.3 \mathrm{mmol}$ ) was suspended in EtOH ( 40 mL )-water ( 20 mL ) and $1 \mathrm{~N} \mathrm{NaOH}(9 \mathrm{~mL})$ was added. The mixture was heated under reflux for 0.5 h and the hot solution was acidified with 1 N HCl $(9.5 \mathrm{~mL})$. The mixture was cooled in an ice bath to crystallize the acid 47 (Table II); IR 1710, 3140, 2100-3000, and 3200-3750 $\mathrm{cm}^{-1}$. Acid 46 (Table II) was prepared in the same way from ester 44.

The following acids were prepared in a similar manner from the esters obtained from the corresponding 2-quinolinamines: compounds 59, 60, 62-65, 72, 73, 80-83, 90, 97-99, 104-110, 112, and 113 with yields in the range $63-96 \%$ (Tables II and III)
$\boldsymbol{N}$-(1 $\boldsymbol{H}$-Tetrazol-5-yl)imidazo[1,2-a $]$ quinoline-2-carboxamide (48) and 8-methoxy- $N$-( $1 H$-tetrazol-5-yl)imidazo-[1,2-a ]quinoline-2-carboxamide (96) were synthesized from the corresponding acids 47 and 81 as for the preparation of 26 (Tables II and III).

2-Chloroethyl Imidazo[1,2-a ]quinoline-2-carboxylate (49) and (2,2-Dimethyl-1,3-dioxolan-4-yl)methyl Imidazo[1,2 a ]quinoline-2-carboxylate (50). Dried acid 47 ( $5 \mathrm{~g}, 23.6 \mathrm{mmol}$ ) was added to $\mathrm{SOCl}_{2}(90 \mathrm{~mL}$ ) containing DMF ( 10 drops) and the mixture was heated under reflux for 3 h . The excess $\mathrm{SOCl}_{2}$ was removed in vacuo by azeotroping with toluene and trituration with $\mathrm{Et}_{2} \mathrm{O}$ to give the crude acid chloride. This solid ( $3 \mathrm{~g}, 13 \mathrm{mmol}$ ) was added to a solution of glycerol acetonide ( $1.8 \mathrm{~g}, 13.6 \mathrm{mmol}$ ) and $\mathrm{NEt}_{3}(1.46 \mathrm{~g}, 14.5 \mathrm{mmol})$ in $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{Cl}_{2}(100 \mathrm{~mL})$ and the mixture obtained was heated under reflux overnight. The solution was then evaporated and the residue was chromatographed ( $\mathrm{SiO}_{2}$; $\mathrm{CHCl}_{3}$ ) to give two products. The less polar material was recrystallized from EtOAc to give the chloroethyl ester 49 (Table II). The more polar material was recrystallized from $\mathrm{CHCl}_{3}-\mathrm{Et}_{2} \mathrm{O}$ to give 50 (Table II).

2,3-Dihydroxypropyl Imidazo[1,2-a ]quinoline-2carboxylate (51). A suspension of glycerol acetonide ester 50 ( $0.45 \mathrm{~g}, 1.38 \mathrm{mmol}$ ) in water ( 90 mL ) containing citric acid monohydrate ( $0.48 \mathrm{~g}, 2.3 \mathrm{mmol}$ ) was heated under reflux for 4 h . After cooling, saturated $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution was added to precipitate the ester 51 (Table II)

Imidazo[1,2-a ]quinoline-2-carboxamide (52). The ester 45 $(0.5 \mathrm{~g}, 2.1 \mathrm{mmol}$ ) was dissolved in $\mathrm{EtOH}(200 \mathrm{~mL})$ and sodium ( $25 \mathrm{mg}, 1.1 \mathrm{mg}$-atom) was added. $\mathrm{NH}_{3}$ gas was bubbled through the solution for 5 h . Stirring was continued for 24 h and then more $\mathrm{NH}_{3}$ gas was bubbled through for 3 h . The solution was concentrated, cooled in an ice bath, acidified with dilute HCl , and then basified with $\mathrm{NaHCO}_{3}$ solution. The solution was extracted with $\mathrm{CHCl}_{3}(2 \times 100 \mathrm{~mL})$, the combined $\mathrm{CHCl}_{3}$ extract was evaporated, and the residue was triturated with $\mathrm{Et}_{2} \mathrm{O}$ to give the
the amide 52 (Table II). Amide 94 (Table III) was prepared in the same way from the ester precursor of acid 81.

Imidazo[1,2-a ]quinoline-2-methanol (53). Ester 45 ( 7.2 g , 30 mmol ) was dissolved in dry THF ( 140 mL ) and stirred while being heated under reflux with $\mathrm{LiBH}_{4}(1.0 \mathrm{~g}, 46 \mathrm{mmol})$ for 20 h . On cooling the solution was acidified with 2 N HCl , stirred for 1 h , basified with saturated $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution, and evaporated. Water ( 500 mL ) was added and the solution was extracted with $\mathrm{CHCl}_{3}(2 \times 200 \mathrm{~mL})$. The combined $\mathrm{CHCl}_{3}$ extract was washed once with water and evaporated. Addition of $\mathrm{EtOAc}_{\mathrm{E}} \mathrm{Et}_{2} \mathrm{O}$ gave the alcohol 53 (Table II). Methanol 91 was prepared in the same way from the ester precursor of acid 81 and likewise 101 from 100 (Table III).

Imidazo[1,2-a ]quinoline-2-carboxaldehyde (54). The alcohol $53(4.2 \mathrm{~g}, 21.2 \mathrm{mmol})$ in $\mathrm{CHCl}_{3}(500 \mathrm{~mL})$ was heated under reflux with activated $\mathrm{MnO}_{2}(16 \mathrm{~g}, 184 \mathrm{mmol})$ for 4 h and then stirred at room temperature overnight. The mixture was filtered through a Celite pad and the pad was washed with $\mathrm{CHCl}_{3}$. The $\mathrm{CHCl}_{3}$ filtrate was evaporated and the residue was triturated with $\mathrm{Et}_{2} \mathrm{O}$ to give the aldehyde 54 (Table II). Aldehyde 92 was prepared in the same way from methanol 91, and likewise 102 from 101 (Table III).

Imidazo[1,2-a ]quinoline-2-methanamine Dihydrochloride (56). The aldehyde $54(2.4 \mathrm{~g}, 12.2 \mathrm{mmol})$ in $\mathrm{EtOH}(60 \mathrm{~mL})$ was treated with a solution of $\mathrm{NH}_{2} \mathrm{OH} \cdot \mathrm{HCl}(1.0 \mathrm{~g}, 14.4 \mathrm{mmol})$ and $\mathrm{NaOAc}(1.35 \mathrm{~g}, 16.5 \mathrm{mmol})$ in water $(20 \mathrm{~mL})$. The mixture was heated under reflux for 2 h and then concentrated under reduced pressure. Water ( 50 mL ) was added to crystallize imidazo[1,2-a]quinoline-2-carboxyaldehyde oxime (55) ( $2.55 \mathrm{~g}, 98 \%$ ), mp $224-226{ }^{\circ} \mathrm{C}$ (from $\mathrm{CHCl}_{3}-\mathrm{MeOH}$ ). A solution of $55(1.0 \mathrm{~g}, 4.74$ mmol ) in 0.25 N methanolic $\mathrm{HCl}(100 \mathrm{~mL})$ was hydrogenated at atmospheric pressure in the presence of $10 \% \mathrm{Pd} / \mathrm{C}(100 \mathrm{mg})$. After 4 h absorption of hydrogen ceased ( 125 mL used), further portion of catalyst ( 200 mg ) was added and hydrogenation was continued at $50^{\circ} \mathrm{C}$ for 2 h ( 62 mL further used). The mixture was then filtered through Celite and evaporated. Trituration with dry $\mathrm{Et}_{2} \mathrm{O}$ gave 56 (Table II). Methanamine 95 was prepared in the same way from aldehyde 92 (Table III).

2 -(1H-Tetrazol-5-yl)imidazo[1,2-a $]$ quinoline (58). The oxime $55(2.0 \mathrm{~g}, 9.5 \mathrm{mmol})$ m $\mathrm{Ac}_{2} \mathrm{O}(20 \mathrm{~mL})$ was stirred while being heated under reflux for 3 h and then allowed to stand overnight at room temperature. The solution was poured into saturated $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution ( 200 mL ) and extracted with $\mathrm{CHCl}_{3}(3 \times 100$ mL ). The combined $\mathrm{CHCl}_{3}$ extract was washed with dilute $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution ( 100 mL ) and water ( 100 mL ) and then evaporated. Trituration with $\mathrm{Et}_{2} \mathrm{O}$ gave imidazo[1,2-a]quinoline-2carbonitrile (57) ( $1.3 \mathrm{~g}, 71 \%$ ) as pale yellow crystals, mp 244-246 ${ }^{\circ} \mathrm{C}$ (from $\left.\mathrm{CHCl}_{3}-\mathrm{Et}_{2} \mathrm{O}\right)$. The nitrile $57(1.0 \mathrm{~g}, 5.2 \mathrm{mmol})$ in DMF ( 50 mL ) was stirred at $40^{\circ} \mathrm{C}$ during the addition of $\mathrm{NH}_{4} \mathrm{Cl}(0.4$ $\mathrm{g}, 7.5 \mathrm{mmol})$ and $\mathrm{NaN}_{3}(0.45 \mathrm{~g}, 6.9 \mathrm{mmol})$. The temperature was raised to $100^{\circ} \mathrm{C}$ and stirring was continued for 16 h . The mixture was treated with $\mathrm{NH}_{4} \mathrm{Cl}(0.2 \mathrm{~g}, 3.75 \mathrm{mmol})$ and $\mathrm{NaN}_{3}(0.225 \mathrm{~g}$, 3.45 mmol ) and stirred for a further 6 h at $120^{\circ} \mathrm{C}$. A final addition of $\mathrm{NH}_{4} \mathrm{Cl}(0.2 \mathrm{~g}, 3.75 \mathrm{mmol})$ and $\mathrm{NaN}_{3}(0.225 \mathrm{~g}, 3.45 \mathrm{mmol})$ was made, and the resultant mixture was stirred for 16 h at $100^{\circ} \mathrm{C}$, then cooled to room temperature, and treated dropwise with water ( 50 mL ) to precipitate 58 (Table II). Compounds 93 and 103 were prepared in the same way from the oximes of aldehydes 92 and 102 (Table III).

5-(Methylsulfonyl)imidazo[1,2-a ]quinoline-2-carboxylic Acid (67). 2-Amino-4-(methylsulfinyl)quinoline was reacted with ethyl bromopyruvate as in the preparation of 45 to give ethyl 5 -(methylsulfinyl)imidazo[1,2-a]quinoline-2-carboxylate. This ester ( $1.5 \mathrm{~g}, 5 \mathrm{mmol}$ ) was stirred in $\mathrm{AcOH}(50 \mathrm{~mL})$ containing $30 \%$ $\mathrm{H}_{2} \mathrm{O}_{2}$ ( $5 \mathrm{~mL}, 44 \mathrm{mmol}$ ) for 1 week at room temperature. The solution was then basified with saturated $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution to give ethyl 5-(methylsulfonyl)imidazo[1,2-a]quinoline-2-carboxylate (66) ( $1.26 \mathrm{~g}, 80 \%$ ), mp $246-247{ }^{\circ} \mathrm{C}$ (from $\mathrm{CHCl}_{3}-\mathrm{Et}_{2} \mathrm{O}$ ). Anal. ( $\mathrm{C}_{15}-$ $\left.\mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{~S}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}, \mathrm{S}$. Hydrolysis of the ester as in the preparation of 47 gave the acid 67 (Table II).

5-(Methylsulfonimidoyl)imidazo[1,2-a ]quinoline-2carboxylic Acid (69). Ethyl 5-(methylsulfonyl)imidazo[1,2-a]quinoline-2-carboxylate ( 66 ) ( $1.5 \mathrm{~g}, 5 \mathrm{mmol}$ ) was stirred in PPA $(40 \mathrm{~mL})$ at $60^{\circ} \mathrm{C}$ during the addition of $\mathrm{NaN}_{3}(400 \mathrm{mg}, 6.2 \mathrm{mmol})$ over a 2 -h period. Stirring and heating were continued for a further 6 h while two further portions of $\mathrm{NaN}_{3}(2 \times 150 \mathrm{mg}, 2$
$\times 2.3 \mathrm{mmol}$ ) were added. The mixture was cooled and poured into ice-water ( 250 mL ), basified with $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution, and extracted with EtOAc $(2 \times 300 \mathrm{~mL})$. The combined EtOAc extract was washed once with water, dried, and evaporated. Trituration with $\mathrm{Et}_{2} \mathrm{O}$ gave ethyl 5-(methylsulfonimidoyl)imidazo $[1,2-a]-$ quinoline-2-carboxylate ( 68 ) ( $1.1 \mathrm{~g}, 70 \%$ ), $\mathrm{mp}^{2} 26-228^{\circ} \mathrm{C}$ (from EtOAc). Anal. $\left(\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$. Hydrolysis of the $\mathrm{e}^{2}$ ter as in the preparation of 47 gave the acid 69 (Table II).

5-Phenylimidazo[1,2-a ]quinoline-2-carboxylic Acid (71). 2-Amino-6-chloro-4-phenylquinoline was reacted with ethyl bromopyruvate as in the preparation of 45 to give ethyl 7 -chloro-5-phenylimidazo[1,2-a]quinoline-2-carboxylate. A mixture of this ester ( $2.1 \mathrm{~g}, 6 \mathrm{mmol}$ ), $\mathrm{NaOAc}(1.0 \mathrm{~g}, 12.2 \mathrm{mmol}), \mathrm{Pd} / \mathrm{C}(5 \%$, 250 mg ), DME ( 75 mL ), and EtOH ( 75 mL ) was stirred under $\mathrm{H}_{2}$ at atmospheric pressure for 2 h . After being filtered through Celite, the solution was evaporated and the residue was dissolved in $\mathrm{CHCl}_{3}(200 \mathrm{~mL})$. After the mixture was washed with water ( 100 mL ), the solvent was removed under reduced pressure and the residue triturated with $\mathrm{Et}_{2} \mathrm{O}$ to give ethyl 5-phenylimidazo-[1,2-a]quinoline-2-carboxylate ( 70 ) ( $1.82 \mathrm{~g}, 98 \%$ ) , mp 144-145 ${ }^{\circ} \mathrm{C}$ (from $\mathrm{CHCl}_{3}-\mathrm{Et}_{2} \mathrm{O}$ ). Anal. ( $\mathrm{C}_{18} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}$ ) H, N; C: calcd, 56.5; found, 57.8. Hydrolysis of the ester as in the preparation of 47 gave the acid 71 (Table II).

Ethyl 4-Cyanoimidazo[1,2-a ]quinoline-2-carboxylate (74). 2-Amino-3-(hydroxymethyl)quinoline was reacted with ethyl bromopyruvate as in the preparation of 45 to give ethyl 4-(hy-droxymethyl)imidazo[1,2-a]quinoline-2-carboxylate. The hydroxymethyl group was converted into a cyano group via the aldehyde and oxime, as in the preparation of 57 from 53 , to give 74 (Table II).

1-Bromoimidazo [1,2-a ]quinoline-2-carboxylic Acid (76). A solution of $\mathrm{Br}_{2}$ in $\mathrm{AcOH}(10 \% \mathrm{w} / \mathrm{v}, 17.6 \mathrm{~mL}, 11 \mathrm{mmol}$ ) was added dropwise to a stirred solution of $45(2.4 \mathrm{~g}, 10 \mathrm{mmol})$ and $\mathrm{NaOAc}(1 \mathrm{~g}, 12.2 \mathrm{mmol})$ in $\mathrm{AcOH}(50 \mathrm{~mL})$. The mixture was warmed on a water bath for 1 h and poured into aqueous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ $(1 \%, 200 \mathrm{~mL})$. The precipitate was filtered and dissolved in $\mathrm{CHCl}_{3}$ $(100 \mathrm{~mL})$. The solution was washed with aqueous $\mathrm{Na}_{2} \mathrm{CO}_{3}$ and evaporated to give ethyl 1-bromoimidazo[1,2-a]quinoline-2carboxylate ( $\mathbf{7 5}$ ) $\left(2.85 \mathrm{~g}, 90 \%\right.$ ), mp $132-133^{\circ} \mathrm{C}$ (from $\mathrm{CHCl}_{3}-\mathrm{Et}_{2} \mathrm{O}$ ). Anal. $\left(\mathrm{C}_{14} \mathrm{H}_{11} \mathrm{BrN}_{2} \mathrm{O}_{2}\right) \mathrm{C}, \mathrm{H}, \mathrm{Br}, \mathrm{N}$. Hydrolysis of 75 as in the preparation of 47 gave 76 (Table II).

5-Chloro-1-methylimidazo[1,2-a ]quinoline-2-carboxylic acid (78) (Table II) was prepared by reacting 2 -amino-4chloroquinoline with ethyl 3 -bromo-2-oxobutanoate as in the synthesis of 45 and hydrolyzing the resulting ester 77 as in the preparation of 47 . Acid 89 (Table III) was prepared in the same way from 2 -amino- 7 -chloroquinoline.

8-(Methylsulfinyl)imidazo[1,2-a ]quinoline-2-carboxylic Acid (85). Ethyl 8-(methylthio)imidazo $[1,2-a]$ quinoline-2carboxylate ( $6 \mathrm{~g}, 21 \mathrm{mmol}$ ), obtained from 2 -amino- 7 -(methylthio) quinoline, was dissolved in $\mathrm{MeOH}(300 \mathrm{~mL})$ and a solution of $\mathrm{NaIO}_{4}(6 \mathrm{~g}, 28 \mathrm{mmol})$ in water ( 30 mL ) was added. The mixture was allowed to stand at room temperature for 65 h and then filtered and the filtrate was evaporated. The residue was dissolved $\mathrm{m} \mathrm{CHCl}_{3}(100 \mathrm{~mL})$ and the organic solution was washed with water and evaporated to give a cream solid, which was chromatographed ( $\mathrm{SiO}_{2} ; 5 \% \mathrm{MeOH}$ in $\mathrm{CHCl}_{3}$ ) to give ethyl 8-(methylsulfinyl)imidazo[ $1,2-a$ ]quinoline-2-carboxylate ( 84 ) $(4.88 \mathrm{~g}, 77 \%), \mathrm{mp}$ $225-227^{\circ} \mathrm{C}$ (from MeOH ). Anal. ( $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$ ) $\mathrm{C}, \mathrm{H}, \mathrm{N}, \mathrm{S}$. Hydrolysis of 84 as in the preparation of 47 gave the acid 85 (Table III).

8-(Methylsulfonyl)imidazo[1,2-a ]quinoline-2-carboxylic Acid (87). The methylthio ester used in the preparation of 84 ( $2.5 \mathrm{~g}, 8.7 \mathrm{mmol}$ ) was suspended in $\mathrm{AcOH}(50 \mathrm{~mL}$ ) and hydrogen peroxide ( $100 \mathrm{w} / \mathrm{v}, 5 \mathrm{~mL}$ ) was added. The mixture was allowed to stand at room temperature for 3 days, then poured into water ( 300 mL ), and extracted with $\mathrm{CHCl}_{3}(2 \times 150 \mathrm{~mL})$. The combined $\mathrm{CHCl}_{3}$ extract was washed with dilute $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution and water and evaporated to give a yellow solid, which was chromatographed ( $\mathrm{SiO}_{2} ; 5 \% \mathrm{MeOH}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give ethyl 8-(methylsulfonyl)imidazo [1,2-a]quinoline-2-carboxylate ( 86 ) ( $1.71 \mathrm{~g}, 62 \%$ ) , mp $249-251{ }^{\circ} \mathrm{C}$, (from $\mathrm{CHCl}_{3}$-EtOAc). Anal. ( $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{~S}$ ) C, H, $\mathrm{N}, \mathrm{S}$. Hydrolysis of 86 as in the preparation of 47 gave the acid 87 (Table III).

8-(Methylsulfonimidoyl)imidazo[1,2-a ]quinoline-2carboxylic acid (88) (Table III) was prepared by hydrolysis of
the corresponding ethyl ester, which was obtained from reaction of 86 with $\mathrm{NaN}_{3}$ as in the synthesis of 68 .
(c) Imidazo[1,2-a ]quinoxalines. 2-Aminoquinoxaline and 2-aminoquinoxaline-3-carboxamide were prepared as described. ${ }^{42}$
2-Amino-6,7-dichloroquinoxaline. A solution of 4,5-di-chloro-1,2-phenylenediamine ( $4 \mathrm{~g}, 22.6 \mathrm{mmol}$ ) in $\mathrm{AcOH}(50 \mathrm{~mL}$ ) was added to a solution of alloxan hydrate ( $3.6 \mathrm{~g}, 22.4 \mathrm{mmol}$ ) and boric acid ( $0.6 \mathrm{~g}, 9.8 \mathrm{mmol}$ ) in $\mathrm{AcOH}(50 \mathrm{~mL})$. The reaction mixture was stirred at room temperature overnight when a brown solid separated. The product was filtered, washed well with water, and dried to give the dichloroalloxazine ( $5.43 \mathrm{~g}, 85 \%$ ), as a yellow crystalline solid, $\mathrm{mp}>370^{\circ} \mathrm{C}$ (from EtOH).

The dichloroalloxazine ( $2 \mathrm{~g}, 7.1 \mathrm{mmol}$ ) was dissolved in concentrated $\mathrm{H}_{2} \mathrm{SO}_{4}(10 \mathrm{~mL})$, heated gradually to $240^{\circ} \mathrm{C}$, and kept at this temperature for 10 min . The mixture was cooled, poured onto ice-water, and basified with aqueous $\mathrm{NaOH}(2 \mathrm{~N}, 100 \mathrm{~mL})$. The aqueous solution was extracted with $\mathrm{Et}_{2} \mathrm{O}(4 \times 100 \mathrm{~mL})$ and the ethereal solution so obtained was washed once with water ( 50 mL ), dried, and evaporated. Trituration with $\mathrm{Et}_{2} \mathrm{O}$ gave 2-amino-6,7-dichloroquinoxaline ( $0.73 \mathrm{~g}, 48 \%$ ), as an orange crystalline solid, mp $220-222{ }^{\circ} \mathrm{C}$ (from EtOH). Anal. ( $\mathrm{C}_{8} \mathrm{H}_{5} \mathrm{Cl}_{2} \mathrm{~N}_{3}$ ) $\mathrm{C}, \mathrm{H}, \mathrm{Cl}, \mathrm{N}$.
Imidazo[1,2-a ]quinoxaline-2-carboxylic acid (114) (Table IV) was prepared by reacting ethyl bromopyruvate with 2aminoquinoxaline as in the synthesis of 45 and then hydrolyzing the ester as in the synthesis of 47 . Compounds 115-117 were prepared in the same way from the corresponding quinoxalines.
$\boldsymbol{N}$-(1H-Tetrazol-5-yl)imidazo[1,2-a ${ }^{\text {- }}$ quinoxaline-2carboxamide (118) (Table IV) was prepared from the acid 114 as for 26.

Imidazo[1,2-a ]quinoxaline-2-methanol (120). Ethyl imidazo[ $1,2-a$ ]quinoxaline- 2 -carboxylate ( $14 \mathrm{~g}, 98 \mathrm{mmol}$ ) obtained in the preparation of 114 was dissolved in dry THF ( 320 mL ) and stirred and heated under reflux for 20 h with $\mathrm{LiBH}_{4}(2 \mathrm{~g}, 92 \mathrm{mmol})$. The mixture was then cooled and poured into aqueous HCl (2 $\mathrm{N}, 200 \mathrm{~mL}$ ). After 1 h the solution was made basic with $\mathrm{Na}_{2} \mathrm{CO}_{3}$ and extracted with EtOAc ( $2 \times 200 \mathrm{~mL}$ ). The combined EtOAc extract was dried and evaporated. Chromatography of the residue ( $\mathrm{SiO}_{2} ; 3 \% \mathrm{MeOH}$ in $\mathrm{CHCl}_{3}$ ) gave 4,5-dihydroimidazo[1,2-a]-quinoxaline-2-methanol (119) (9.56 g, $82 \%$ ), mp 172-174 ${ }^{\circ} \mathrm{C}$ (from EtOAc). A mixture of $119(2.3 \mathrm{~g}, 11.5 \mathrm{mmol})$ in $\mathrm{CHCl}_{3}(300 \mathrm{~mL})$ and activated $\mathrm{MnO}_{2}(4 \mathrm{~g}, 46 \mathrm{mmol})$ was stirred vigorously for 2 h at room temperature and then filtered through Celite. The filtrate was evaporated to a small volume and the solution was cooled in an ice bath to crystallize 120 (Table IV).

Imidazo[1,2-a ]quinoxaline-2-carboxaldehyde (121). A solution of $119(2.5 \mathrm{~g}, 12.5 \mathrm{mmol})$ in $\mathrm{CHCl}_{3}(500 \mathrm{~mL})$ was heated under reflux for 3 h with activated $\mathrm{MnO}_{2}(12 \mathrm{~g}, 138 \mathrm{mmol})$. The mixture was then filtered through Celite and the filtrate was evaporated. Trituration with $\mathrm{Et}_{2} \mathrm{O}$ gave 121 (Table IV).

2-( $1 \boldsymbol{H}$-Tetrazol-5-yl)imidazo[1,2-a ]quinoxaline (122) (Table IV) was prepared from 121 as in the synthesis of 58 from 54.
(d) Imidazo[1,2-a ]quinoxalin-4(5H)-ones. Ethyl 4,5-Di-hydro-4-oxoimidazo [1,2-a ]quinoxaline-2-carboxylate (123). 2-Amino-3-chloroquinoxaline ${ }^{16}(6.1 \mathrm{~g}, 34 \mathrm{mmol})$ and ethyl bromopyruvate ( $8.0 \mathrm{~g}, 41 \mathrm{mmol}$ ) in dimethoxyethane ( 120 mL ) were stirred at room temperature overnight. A small amount of yellow crystalline solid was filtered and the filtrate was stood at room temperature. Over the next 2 weeks four further crops of quaternary salt were filtered (total yield $7.81 \mathrm{~g}, 61 \%$ ); IR $1763 \mathrm{~cm}^{-1}$ (ester carbonyl). The quaternary salt ( $6.5 \mathrm{~g}, 17.5 \mathrm{mmol}$ ) was heated under reflux in EtOH ( 500 mL ) for 2 h to give a clear yellow solution. The EtOH solution was concentrated to ca. 50 mL to crystallize 123 ( $3.75 \mathrm{~g}, 85 \%$; $52 \%$ overall), $\mathrm{mp} 292-293^{\circ} \mathrm{C}$ (from EtOH); IR 743, 1261, 1274, 1538, 1701 (br), and $3135 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{Me}_{2} \mathrm{SO}-d_{6}\right) \delta 1.33\left(3 \mathrm{H}, \mathrm{t}, \mathrm{CH}_{3}\right), 3.40(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 4-\mathrm{OH}), 4.30$ $\left(2 \mathrm{H}, \mathrm{q}, \mathrm{CH}_{2}\right), 7.0-7.45(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} \mathrm{H}), 8.05(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H})$, and 9.05 ( $1 \mathrm{H}, \mathrm{s}, 1-\mathrm{H}$ ). IR and NMR suggest it may exist in the tautomeric 4-OH form. Anal. ( $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}_{3}$ ) C, $\mathrm{H}, \mathrm{N}$. Hydrolysis of 123 as in the preparation of 47 gave the acid 127 (Table V).

5-Alkyl-4,5-dihydro-4-oxoimidazo [1,2-a ]quinoxaline-2carboxylic Acids (128-131). A solution of 123 in DMF was treated with NaH ( 1 equiv) followed by an alkyl iodide ( 1.1 equiv) to yield the corresponding ethyl 5 -alkyl-4,5-dihydro-4-oxo-imidazo[1,2-a]quimoxaline-2-carboxylate, for example 5 -ethyl (124), 5 -propyl (125), and 5 -benzyl (126) (Table V). Hydrolysis of these
esters as in the preparation of 47 gave the corresponding acids 128-131 (Table V).

7,8-Dichloro-4,5-dihydro-4-oxoimidazo[1,2-a]-quinoxaline-2-carboxylic Acid (133). 2,3,6,7-Tetrachloroquinoxaline ${ }^{43}(3.46 \mathrm{~g}, 13 \mathrm{mmol})$ was added to $\mathrm{EtOH}(30 \mathrm{~mL})$ and the mixture was saturated with $\mathrm{NH}_{3}$ gas at $0^{\circ} \mathrm{C}$. The reaction mixture was shaken under pressure ( 60 psi ) at $80^{\circ} \mathrm{C}$ overnight. The solution was then cooled, evaporated, and triturated with water to give a buff crystalline solid, which was purified by chromatography ( $\mathrm{SiO}_{2} ; 50 \% \mathrm{EtOAc}$ in $40-60^{\circ} \mathrm{C}$ petroleum ether) to give 2-amino-6,7-dichloro-3-ethoxyquinoxaline ( $1.36 \mathrm{~g}, 41 \%$ ) as a pale yellow crystalline solid: $\mathrm{mp} 191-193^{\circ} \mathrm{C}$ (from Et-$\mathrm{OAc}-\mathrm{Et}_{2} \mathrm{O}$ ); IR 882, 1033, 1231, 1281, 1471, 1660, and $3500 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} N M R\left(\mathrm{Me}_{2} \mathrm{SO}-d_{6}\right) \delta 1.39\left(3 \mathrm{H}, \mathrm{t}, \mathrm{CH}_{3}\right), 4.43\left(2 \mathrm{H}, \mathrm{q}, \mathrm{CH}_{2}\right), 7.10$ ( 2 H , br s, $\mathrm{NH}_{2}$ ), and $7.52,7.63(2 \times 1 \mathrm{H}, 2 \mathrm{~s}, 5-$ and $8-\mathrm{H})$. Anal. $\left(\mathrm{C}_{10} \mathrm{H}_{9} \mathrm{Cl}_{2} \mathrm{~N}_{3} \mathrm{O}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$. A solution of this amine $(1.0 \mathrm{~g}, 3.9 \mathrm{mmol})$ and ethyl bromopyruvate ( $1.0 \mathrm{~g}, 5.1 \mathrm{mmol}$ ) in dimethoxyethane $(50 \mathrm{~mL})$ was stirred at room temperature for 1 week. A pinkish white solid was filtered and chromatographed $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}\right)$ to give ethyl 7,8-dichloro-4-ethoxyimidazo[1,2-a]quinoxaline-2carboxylate (132) ( $0.15 \mathrm{~g}, 11 \%$ ): mp $256-258^{\circ} \mathrm{C}$ (from EtOH$\mathrm{Et}_{2} \mathrm{O}$ ); IR $1700,3130 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $\mathrm{Me}_{2} \mathrm{SO}-d_{6}$ ) $\delta 1.37(6 \mathrm{H}, \mathrm{t}$, $\left.2 \mathrm{CH}_{3}\right), 4.40\left(4 \mathrm{H}, \mathrm{q}, 2 \mathrm{CH}_{2}\right), 7.82(1 \mathrm{H}, \mathrm{s}, 6-\mathrm{H}), 8.68(1 \mathrm{H}, \mathrm{s}, 9-\mathrm{H})$, and $9.26(1 \mathrm{H}, \mathrm{s}, 1-\mathrm{H})$. Anal. $\left(\mathrm{C}_{15} \mathrm{H}_{13} \mathrm{Cl}_{3} \mathrm{~N}_{3} \mathrm{O}_{3}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$. To the ester $132(0.14 \mathrm{~g}, 0.39 \mathrm{mmol})$ suspended in $\mathrm{EtOH}(10 \mathrm{~mL})$ was added aqueous $\mathrm{NaOH}(1 \mathrm{~N}, 3 \mathrm{~mL}$ ) and water ( 20 mL ). The mixture so obtained was heated under reflux overnight and then cooled and acidified with concentrated HCl to $\mathrm{pH} 2-3$ to precipitate 133 (Table V); IR 1700,3130 , and $3150-3410 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{Me}_{2} \mathrm{SO}-d_{6}\right) \delta 7.42(1 \mathrm{H}, \mathrm{s}, 6-\mathrm{H}), 8.54(1 \mathrm{H}, \mathrm{s}, 9-\mathrm{H})$, and 9.07 ( $1 \mathrm{H}, \mathrm{s}, 1-\mathrm{H}$ ).

5-Ethyl-4,5-dihydro-4-oxoimidazo[1,2-a ]quinoxaline-2methanol (134) (Table V) was prepared from 123 as for the synthesis of 53 .

5-Ethyl-4,5-dihydro-4-oxoimidazo[1,2-a ]quinoxaline-2carboxaldehyde (135) (Table V) was prepared from 134 as for the synthesis of 54 .

5-Ethyl-4,5-dihydro-4-oxo- $\boldsymbol{N}$-(1H-tetrazol-5-yl)imidazo-[1,2-a ]quinoxaline-2-carboxamide (136) (Table V) was prepared from 129 as for the synthesis of 26 .
(e) Imidazo[1,2-a ]quinazolin-5(4H)-ones. 4-Ethyl-4,5-dihydro-5-oxoimidazo[1,2-a ]quinazoline-2-carboxylic Acid (138). 2-Amino-3-ethylquinazolin- $4(3 H)$-one ${ }^{44}(2.90 \mathrm{~g}, 15.3 \mathrm{mmol})$ and ethyl bromopyruvate ( $4.5 \mathrm{~g}, 23 \mathrm{mmol}$ ) were dissolved in dimethoxyethane ( 50 mL ), and the mixture was stirred for 3 h . $\mathrm{EtOH}(50 \mathrm{~mL})$ was added and the mixture was heated under reflux for 3 h and then allowed to stand at room temperature for 2 days to crystallize ethyl 4-ethyl-4,5-dihydro-5-oxoimidazo[1,2-a]-quinazoline-2-carboxylate (137) (Table VI); IR 1672, 1713, and $3135 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.40\left(6 \mathrm{H}, \mathrm{t}, 2 \mathrm{CH}_{3}\right), 4.38(2 \mathrm{H}, \mathrm{q}$, $\left.\mathrm{CH}_{2}\right), 4.35\left(2 \mathrm{H}, \mathrm{q}, \mathrm{CH}_{2}\right), 7.3-8.0(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} \mathrm{H}), 8.09(1 \mathrm{H}, \mathrm{s}, 1-\mathrm{H})$, and $8.30(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H})$. Hydrolysis of 137 as in the preparation of 47 gave the acid 138 (Table VI).
(f) Pyrrolo[1,2-a ]quinoxalin-4(5H)-ones. 1-Substituted benzimidazoles were prepared ${ }^{45}$ by treating benzimidazoles with alkyl or alkenyl bromo or iodo compounds in the presence of base. Chromatography $\left(\mathrm{Al}_{2} \mathrm{O}_{3} ; \mathrm{CHCl}_{3}\right)$ gave the 1 -substituted benzimidazoles as crude oils, which were used without distillation. 1-Phenylbenzimidazole ${ }^{46}$ was prepared from $N$-phenyl-ophenylenediamine by cyclization with formic acid.

1-[(Ethoxycarbonyl)methyl]-3-ethylbenzimidazolium Bromide (140). Crude 1-ethylbenzimidazole ( $30.7 \mathrm{~g}, 0.21 \mathrm{~mol}$ ) was dissolved in $\mathrm{Et}_{2} \mathrm{O}(300 \mathrm{~mL})$ and ethyl bromoacetate ( 39 g , 0.234 mol ) was added. After the mixture was allowed to stand at room temperature for 3 days, a crystalline precipitate of 140 was formed ( $53.0 \mathrm{~g}, 57 \%$ ): mp 119-121 ${ }^{\circ} \mathrm{C}$ (from $\mathrm{MeOH}-\mathrm{Et}_{2} \mathrm{O}$ ); IR $760,1231,1570,1753,3415$, and $3490 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right)$
(42) Gowenlock, A. H.; Newbold, G. T.; Spring, F. S. J. Chem. Soc. 1945, 622.
(43) Cheeseman, G. W. H. J. Chem. Soc. 1962, 1170.
(44) Grout, R. J.; Partridge, M. W. J. Chem. Soc. 1960, 3540.
(45) Pozkorzskii, A. F.; Simonov, A. M. Zh. Obshch. Khim. 1963, 33, 179.
(46) Phillips, M. A. J. Chem. Soc. 1929, 2823.
$\delta 1.27\left(3 \mathrm{H}, \mathrm{t}\right.$, ester $\left.\mathrm{CH}_{3}\right), 1.70\left(3 \mathrm{H}, \mathrm{t}, 3-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.26(2 \mathrm{H}, \mathrm{q}$, ester $\left.\mathrm{CH}_{2}\right), 4.60\left(2 \mathrm{H}, \mathrm{q}, 3-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 5.65\left(2 \mathrm{H}, \mathrm{s}, 1-\mathrm{CH}_{2}\right), 7.4-8.0$ $(4 \mathrm{H}, \mathrm{m}, \mathrm{Ar} \mathrm{H})$, and $10.93(1 \mathrm{H}, \mathrm{s}, 2-\mathrm{H})$. Anal. $\left(\mathrm{C}_{13} \mathrm{H}_{17} \mathrm{BrN}_{2} \mathrm{O}_{2} \cdot \mathrm{H}_{2} \mathrm{O}\right)$ C, H, Br, N.

Ethyl 5-Ethyl-4,5-dihydro-4-oxopyrrolo[1,2-a ]-quinoxaline-2-carboxylate (144). The quaternary salt 140 (20.0 $\mathrm{g}, 64 \mathrm{mmol})$ was dissolved in DMF ( 100 mL ), and $\mathrm{NEt}_{3}(7.4 \mathrm{~g}$, 73 mmol ) and ethyl propiolate ( $8.05 \mathrm{~g}, 83 \mathrm{mmol}$ ) were added. The mixture was allowed to stand at room temperature for 3 days, and then EtOAc $(200 \mathrm{~mL})$ and water ( 200 mL ) were added. The EtOAc layer was separated and the aqueous layer was further extracted with EtOAc $(3 \times 100 \mathrm{~mL})$. The combined EtOAc layer was washed with water ( 200 mL ), dried, and evaporated. The reddish oil formed was triturated with $\mathrm{Et}_{2} \mathrm{O}$ to give 144 ( 9.1 g , $50 \%$ ) as buff needles: $\mathrm{mp} 184-186^{\circ} \mathrm{C}\left(\right.$ from $\left.\mathrm{Et}_{2} \mathrm{O}\right)$; IR 739, 747, $759,1260,1276,1653,1714$, and $3125 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta$ $1.32\left(3 \mathrm{H}, \mathrm{t}, 5-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.36\left(3 \mathrm{H}\right.$, t, ester $\left.\mathrm{CH}_{3}\right), 4.22(2 \mathrm{H}, \mathrm{q}$, $\left.5-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.28\left(2 \mathrm{H}, \mathrm{q}\right.$, ester $\left.\mathrm{CH}_{2}\right), 7.0-7.4(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} \mathrm{H}), 7.45$ $(1 \mathrm{H}, \mathrm{d}, J=1.5 \mathrm{~Hz}, 3-\mathrm{H}), 7.5-7.75(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}), 8.05(1 \mathrm{H}, \mathrm{d} J$ $=1.5 \mathrm{~Hz}, 1-\mathrm{H})$. Anal. $\left(\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{3}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$. Esters 142 and 143 (Table VII) were prepared in the same way from the corresponding salts.

5-Ethyl-4,5-dihydro-4-oxopyrrolo[1,2-a ]quinoxaline-2carboxylic Acid (146). The ester 144 ( $6.5 \mathrm{~g}, 23 \mathrm{mmol}$ ) was suspended in $\mathrm{EtOH}(100 \mathrm{~mL})$ and heated on a steam bath. A solution of $\mathrm{NaOH}(2.0 \mathrm{~g}, 50 \mathrm{mmol})$ in water $(200 \mathrm{~mL})$ was added and the mixture was heated until a clear solution was obtained. The solution was filtered hot and acidified to $\mathrm{pH} 2-3$ (concentrated HCl ). On cooling in an ice bath, crystallization gave 146 (Table VII); IR 738, 765, 1283, 1311, 1654, 1696, and $3140 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{Me}_{2} \mathrm{SO}-d_{6}\right) \delta 1.22\left(3 \mathrm{H}, \mathrm{t}, \mathrm{CH}_{3}\right), 4.19\left(2 \mathrm{H}, \mathrm{q}, \mathrm{CH}_{2}\right), 7.21(1 \mathrm{H}$, $\mathrm{d}, J=1 \mathrm{~Hz}, 3-\mathrm{H}), 7.1-7.6(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} \mathrm{H}), 8.20(1 \mathrm{H}, \mathrm{dd}, 9-\mathrm{H})$, and $8.60(1 \mathrm{H}, \mathrm{d}, J=1 \mathrm{~Hz}, 1-\mathrm{H})$. Acids 145 and 147-157 (Table VII) were prepared in the same way from the corresponding esters.

5-Ethyl-4,5-dihydro-4-oxopyrrolo[1,2-a ]quinoxaline-2methanol (158) (Table VII) was prepared from 150 as for the synthesis of 53 .

5-Ethyl-4,5-dihydro-4-oxopyrrolo[1,2-a ]quinoxaline-2carboxaldehyde (159) (Table VII) was prepared from 158 as for the synthesis of 54 .

5-Ethyl-2-(1H-tetrazol-5-yl)pyrrolo[1,2-a ${ }^{\text {- }}$ quinoxalin-4( $5 H$ )-one ( 160 ) (Table VII) was prepared from 159 as in the synthesis of 58 from 54.

5-Ethyl-4,5-dihydro-4-oxo- $\boldsymbol{N}$-(1H-tetrazol-5-yl) pyrrolo-[1,2-a ]quinoxaline-2-carboxamide (161) (Table VII) was prepared from 146 as for the synthesis of 26.

Dimethyl 5-Ethyl-4,5-dihydro-4-oxopyrrolo[1,2-a ]-quinoxaline-2,3-dicarboxylate (162). The quaternary salt 140 ( $2.7 \mathrm{~g}, 8.6 \mathrm{mmol}$ ) was dissolved in DMF ( 50 mL ), and $\mathrm{NEt}_{3}(1.0$ $\mathrm{g}, 10 \mathrm{mmol}$ ) and dimethyl acetylenedicarboxylate ( $1.5 \mathrm{~g}, 10.5$ mmol ) were added. The mixture was warmed on a water bath for 15 min and then allowed to stand at room temperature overnight. Workup as for 144 gave $162(1.4 \mathrm{~g}, 50 \%)$ as buff crystals: $\mathrm{mp} 198-200^{\circ} \mathrm{C}$ (from $\mathrm{Et}_{2} \mathrm{O}$ ); IR 742, 1261, 1278, 1660, 1722, 1749, and $3160 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.31\left(3 \mathrm{H}, \mathrm{t}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 3.80$ $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 3.97\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 4.21\left(2 \mathrm{H}, \mathrm{q}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.0-7.4$ $(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} \mathrm{H}), 7.5-7.8(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H})$, and $8.02(1 \mathrm{H}, \mathrm{s}, 1-\mathrm{H})$. Anal. $\left(\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{5}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

5-Ethyl-4,5-dihydro-3-(methoxycarbonyl)-4-oxopyrrolo-[1,2-a ]quinoxaline-2-carboxylic Acid (163). The diester 162 ( $0.85 \mathrm{~g}, 2.6 \mathrm{mmol}$ ) was suspended in $\mathrm{EtOH}(30 \mathrm{~mL})$ and a solution of $\mathrm{NaOH}(0.5 \mathrm{~g}, 12.5 \mathrm{mmol})$ in water $(10 \mathrm{~mL})$ was added. The mixture was heated on a water bath for 2 h and then filtered and acidified (concentrated HCl ) to $\mathrm{pH} 2-3$ to precipitate 163 (Table VII).
(g) Pyrazolo[2,3-a ]quinoxalin-4(5H)-ones. 4,5-Dihydro-4-oxopyrazolo[2,3-a ]quinoxaline-2-carboxylic Acid (166). A suspension of the diacid ${ }^{18} 165(7.65 \mathrm{~g}, 0.027 \mathrm{~mol})$, platinum oxide ( 200 mg ), and $10 \% \mathrm{Pd} / \mathrm{C}(200 \mathrm{mg})$ in aqueous $\mathrm{NaOH}(10 \%)$ was stirred under $\mathrm{H}_{2}$ ( 4 atm ) for 6 h . The mixture was filtered through Celite and acidified with concentrated HCl to $\mathrm{pH} 2-3$ to give the acid 166 (Table VIII).

Methyl 4,5-Dihydro-4-oxopyrazolo[2,3-a ]quinoxaline-2carboxylate (167). A suspension of acid $166(3.5 \mathrm{~g}, 15.3 \mathrm{mmol})$ in methanolic $\mathrm{HCl}(5 \%, 250 \mathrm{~mL})$ was heated on a water bath for 3 h . On cooling, the ester 167 ( $3.4 \mathrm{~g}, 92 \%$ ), mp $329-331^{\circ} \mathrm{C}$,
crystallized as white needles (from EtOAc). Anal. ( $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{O}_{3}$ ) C, H, N.

Methyl 4,5-Dihydro-5-methyl-4-oxopyrazolo[2,3-a ]-quinoxaline-2-carboxylate (168). A solution of ester 167 (1.1 $\mathrm{g}, 4.5 \mathrm{mmol}$ ), NaH ( $60 \%$ dispersion in oil; $0.4 \mathrm{~g}, 10 \mathrm{mmol}$ ), and MeI ( $1.14 \mathrm{~g}, 8.0 \mathrm{mmol}$ ) in DMF ( 25 mL ) was stirred at room temperature for 3 h and cooled, after which water ( 100 mL ) and EtOAc ( 150 mL ) were added. The EtOAc layer was separated, washed $\left(\mathrm{H}_{2} \mathrm{O}\right)$, and evaporated to give $168(0.75 \mathrm{~g}, 71 \%)$ : mp $240-242^{\circ} \mathrm{C}$ as colorless needles (from $\mathrm{Et}_{2} \mathrm{O}$ ); IR 742, 1272, 1348, 1660,1725 , and $3140 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 3.70\left(3 \mathrm{H}, \mathrm{s}, 5-\mathrm{CH}_{3}\right)$, $3.97\left(3 \mathrm{H}, \mathrm{s}\right.$, ester $\left.\mathrm{CH}_{3}\right), 7.1-7.6(3 \mathrm{H}, \mathrm{m}, \operatorname{ArH}), 7.61(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H})$, and 8.3-8.6 $(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H})$. Anal. $\left(\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}_{3}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

Use of EtI in place of MeI gave methyl 4,5-dihydro-5-ethyl-4-oxopyrazolo[2,3-a]quinoxaline-2-carboxylate (169), mp 177-178 ${ }^{\circ} \mathrm{C}$ (from $\mathrm{Et}_{2} \mathrm{O}$ ) [anal. $\left(\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}_{3}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$ ], but when PrI was employed, the aqueous workup was carried out at $50^{\circ} \mathrm{C}$ to hydrolyze the intermediate ester to acid 172 (Table VIII). Acids 170 and 171 were obtained from esters 168 and 169 as in the preparation of 47 .
(h) Imidazo[1,5-a ]quinoxalinones, Triazolo[1,5-a ]quinoxalinones, and Triazolo[1,5-a]benzoxazinones. Methyl 4,5-Dihydro-4-oxoimidazo[1,5-a ]quinoxaline-3-carboxylate (174). A solution of diamine $173^{10,25}(1.0 \mathrm{~g}, 4.3 \mathrm{mmol})$ and $p-\mathrm{TsOH}$ ( 5 mg ) in triethyl orthoformate ( $5 \mathrm{~g}, 34 \mathrm{mmol}$ ) was refluxed for 2 h , cooled, and diluted with $\mathrm{Et}_{2} \mathrm{O}$ to precipitate a red solid. This was filtered and chromatographed twice ( $\mathrm{SiO}_{2} ; \mathrm{EtOAc}$ and $5 \%$ $\mathrm{MeOH}-\mathrm{CHCl}_{3}$ ) to give $174(0.81 \mathrm{~g}, 85 \%): \operatorname{mp} 270-271{ }^{\circ} \mathrm{C}$ (Et$\mathrm{OAc}_{\mathrm{CHCl}}^{3}$; ; IR $1445,1485,1520,1565,1630,1680,1735,3000$, 3110,3270 , and $3520 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{Me}_{2} \mathrm{SO}-d_{6}$ ) $\delta 3.8(3 \mathrm{H}$, s, ester $\left.\mathrm{CH}_{3}\right)$, 7.1-7.4 (3 H, m, Ar H), 8.1-8.4 ( $1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}$ ), $9.0(1 \mathrm{H}, \mathrm{s}$, 1-H), $11.3(1 \mathrm{H}$, br s, $5-\mathrm{H})$. Anal. $\left(\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{O}_{3}\right.$ ) C, H, N.

Methyl 4,5-Dihydro-1-methyl-4-oxoimidazo[1,5-a]-quinoxaline-3-carboxylate (175). This was prepared in the same way as 174 but with triethyl orthoacetate (yield $62 \%$ ): mp 255-252 ${ }^{\circ} \mathrm{C}$; IR $1560,1620,1705,1735,2920,3000,3100$, and $3450 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 2.85\left(3 \mathrm{H}, \mathrm{s}, 1-\mathrm{CH}_{3}\right), 3.75\left(3 \mathrm{H}, \mathrm{s}\right.$, ester $\left.\mathrm{CH}_{3}\right)$, 7.3-8.0 ( $4 \mathrm{H}, \mathrm{m}$, Ar H), $11.6(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H})$. Anal. $\left(\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}_{3}\right)$ C, H, N.

4,5-Dihydro-4-oxoimidazo[1,5-a ]quinoxaline-3-carboxylic Acid (178). A suspension of 174 ( 1 g ) in aqueous $\mathrm{EtOH}(50 \%$, 20 mL ) containing $\mathrm{KOH}(1 \mathrm{~g})$ was refluxed until clear, cooled, acidified (concentrated HCl ), and filtered to give 178 (Table IX).

4,5-Dihydro-1-methyl-4-oxoimidazo[1,5-a ]quinoxaline-3carboxylic acid (179) (Table IX) was prepared from 175 in the same way as 178 .

4,5-Dihydro-1-ethyl-4-oxoimidazo[1,5-a ]quinoxaline-3carboxylic Acid (180). A suspension of $176^{9}(100 \mathrm{mg})$ in aqueous MeOH ( $50 \% ; 10 \mathrm{~mL}$ ) containing $\mathrm{NaOH}(100 \mathrm{mg})$ was refluxed for 2 h , cooled, and acidified to give 180 (Table IX).

4,5-Dihydro-1-benzyl-4-oxoimidazo[1,5-a ]quinoxaline-3carboxylic acid (181) (Table IX) was prepared from $177^{9}$ in the same way as 180 .

Methyl 4-Oxo[1,2,3]triazolo[3,4-c][1,4]benzoxazine-3carboxylate (183). A solution of $182^{21}(1.2 \mathrm{~g}, 5 \mathrm{mmol})$ in dioxane ( 5 mL ) and $\mathrm{Et}_{2} \mathrm{O}\left(10 \mathrm{~mL}\right.$ ) at $0^{\circ} \mathrm{C}$ was treated with $\mathrm{Cl}_{3} \mathrm{CCO}_{2} \mathrm{H}$ $(300 \mathrm{mg})$ and amyl nitrite ( 600 mg ), left overnight at $0^{\circ} \mathrm{C}$, diluted with EtOH , and filtered. The filtrate was evaporated and the residue chromatographed $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}\right)$ to give $183(800 \mathrm{mg})$ (Table IX): IR $1510,1610,1620,1740,1770$, and $3100 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 3.9\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 7.5(3 \mathrm{H}, \mathrm{m}, \operatorname{Ar~H}), 8.3(1 \mathrm{H}$, d, $9-\mathrm{H}$ ).

Methyl 4,5-dihydro-4-oxo[1,2,3]triazolo[1,5-a]-quinoxaline-3-carboxylate (184) was prepared from $173^{10,25}$ in the same way as 183 (yield $60 \%$ ): mp $232-234^{\circ} \mathrm{C}$ (EtOAc$\mathrm{CHCl}_{3}$ ); IR 1450, 1495, 1550, 1630, 1700, 1735, and $3320 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{Me}_{2} \mathrm{SO}-d_{6}\right) \delta 3.95\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 7.3-7.7(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} \mathrm{H})$, $8.4(1 \mathrm{H}$, br d, $9-\mathrm{H})$. Anal. $\left(\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{~N}_{4} \mathrm{O}_{3}\right) \mathrm{C}, \mathrm{H}$; N : calcd, 22.9; found, 24.6.

4,5-Dihydro-4-oxo[1,2,3]triazolo[1,5-a ]quinoxaline-3carboxylic acid (185) (Table IX) was prepared by refluxing a suspension of 184 and $\mathrm{NaHCO}_{3}$ in aqueous MeOH followed by precipitation with concentrated HCl .
(i) Miscellaneous Ring Systems. Ethyl Pyrrolo[1,2-a ]-quinoline-2-carboxylate (187). To a refluxing solution of
quinaldine ( $9 \mathrm{~g}, 63 \mathrm{mmol}$ ) in $\mathrm{EtOH}(100 \mathrm{~mL})$ was added over 0.5 h ethyl bromopyruvate ( $13 \mathrm{~g}, 67 \mathrm{mmol}$ ) in EtOH ( 50 mL ). Heating was continued for a further 1.5 h , the solvent was evaporated, and the residue was partitioned between dilute $\mathrm{HCl}-\mathrm{EtOAc}$. The organic layer was separated, washed with dilute $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution and water, and evaporated to give a deep red oil, which was chromatographed $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}\right)$ to give 187 as lemon needles ( 3.41 $\mathrm{g}, 23 \%), \operatorname{mp} 77-78^{\circ} \mathrm{C}\left(\mathrm{Et}_{2} \mathrm{O}\right.$; petroleum ether). Anal. $\left(\mathrm{C}_{15} \mathrm{H}_{13} \mathrm{NO}_{2}\right)$ C, H, N.

Pyrrolo[1,2-a ]quinoline-2-carboxylic Acid (188). A solution of $187(1.2 \mathrm{~g})$ in $\mathrm{EtOH}(10 \mathrm{~mL})$ and aqueous $\mathrm{NaOH}(1 \mathrm{~N}, 1 \mathrm{~mL})$ was refluxed for 4 h , cooled, and acidified (concentrated HCl ) to give 188 (Table IX).

5-Oxo-5-phenylimidazo[3,4-b][1,2,4]benzothiadiazine-2carboxylic Acid (195). Ethyl bromopyruvate ( 750 mg ) was added to a solution of 1-oxo-1-phenylbenzothiadiazin-3-amine (192) ${ }^{47}$ ( 800 mg ) in THF ( 40 mL ) at $0^{\circ} \mathrm{C}$. After the mixture was allowed to stand at room temperature overnight, more ethyl bromopyruvate ( 300 mg ) was added and 4 h later the solvent was evaporated. The residue was refluxed in EtOH for 36 h , and after evaporation of the solvent, the residue was chromatographed $\left(\mathrm{SiO}_{2}\right.$; $\left.\mathrm{CHCl}_{3}-\mathrm{EtOH}\right)$ to give ethyl 5-oxo-5-phenylimidazo[3,4-b][1,2,4] benzothiadiazine-2-carboxylate (194) ( $600 \mathrm{mg}, 47 \%$ ), mp $167-169{ }^{\circ} \mathrm{C}$ (from $\mathrm{CHCl}_{3}-\mathrm{EtOAc}$ ). Anal. ( $\left.\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}\right) \mathrm{H}, \mathrm{N}$, S ; C : calcd, 53.6 ; found 53.1. Hydrolysis of 194 as in the preparation of 188 gave the acid 195 (Table X).

5-Methyl-5-oxoimidazo[3,4-b][1,2,4]ben zothiadiazine-2carboxylic acid (197) (Table X) was prepared in the same way as 195 with 1-methyl-1-oxobenzothiadiazin- 3 -amine as the starting material via ethyl 5 -methyl- 5 -oxoimidazo $[3,4-b][1,2,4]$ benzo-thiadiazine-2-carboxylate (196), mp $270{ }^{\circ} \mathrm{C}$. Anal. ( $\mathrm{C}_{13} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ ) H, N; C: calcd, 53.1; found, 53.6.

Ethyl Imidazo[2,1-b ]benzothiazole-2-carboxylates (198, 200, 202, 204). By quaternizing the corresponding 2-benzothiazolamine ${ }^{48}$ with ethyl bromopyruvate in DME, evaporating the solvent, and refluxing the residue in EtOH , there were obtained after chromatography $\left(\mathrm{SiO}_{2}\right)$ the following esters: unsubstituted (198), mp 142-143 ${ }^{\circ} \mathrm{C}$ (lit. ${ }^{25} \mathrm{mp} 147-148{ }^{\circ} \mathrm{C}$ ) [anal. $\left.\left(\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}, \mathrm{S}\right] ; 7$-methoxy (200), mp 133-136 ${ }^{\circ} \mathrm{C}$ [anal. $\left(\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}\right.$ ) C, H, N, S]; 6,7-dimethyl (202), mp 176-178 ${ }^{\circ} \mathrm{C}$ [anal. ( $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}$ ) C, H, N, S]; 5-methoxy (204), mp 145-148 ${ }^{\circ} \mathrm{C}$ [anal. $\left(\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}, \mathrm{S}$ ].

Imidazo[2,1-b]benzothiazole-2-carboxylic acids (199, 201, 203, 205) (Table X) were obtained by hydrolysis of the corresponding esters as in the preparation of 188.

Registry No. 1, 15826-37-6; 2, 58761-87-8; 3, 65565-73-3; 3 (acid chloride), 65565-75-5; 11, 65565-72-2; 12, 76577-75-8; 13, 76577-76-9; 14, 76577-77-0; 15, 76577-78-1; 16, 76577-79-2; 17, 82296-38-6; 18, 82296-36-4; 19, 65565-85-7; 19 (acid chloride), 82296-30-8; 20, 82296-10-4; 21, 82296-29-5; 22, 82296-32-0; 23, 82296-34-2; 24, 113508-03-5; 25, 113508-04-6; 26, 113508-05-7; 27, 113508-06-8; 28, 113508-07-9; 29, 65565-82-4; 30, 65565-80-2; 31, 65565-83-5; 32, 65565-81-3; 33, 65565-84-6; 34, 70917-73-6; 35, 70917-68-9; 36, 70917-72-5; 37, 70917-77-0; 38, 113508-08-0; 39, 113508-09-1; 43, 62235-41-0; 43 (2-amino), 113508-15-9; 44, 113508-14-8; 45, $68050-48-6 ; 45 \cdot \mathrm{HBr}, 113508-16-0 ; 46,113508-17-1 ; 46 \cdot 2 \mathrm{HCl}$, 113508-63-7; 47, 68050-43-1; 47 (acid chloride), 113508-54-6; 48, 76577-85-0; 49, 113508-18-2; 50, 68069-09-0; 51, 68050-41-9; 52, $76577-86-1 ; 53,76577-48-5 ; 54,76577-49-6 ; 55,76577-50-9 ; 56$, $113508-19-3 ; 56 \cdot 2 \mathrm{HCl}, 113508-64-8 ; 57,76577-51-0 ; 58,76577-52-1$; 59, 66491-26-7; 59 (ethyl ester), 66491-09-6; 60, 68050-47-5; 60 (ethyl ester), 68050-10-2; 61, 67817-46-3; 62, 113508-20-6; 62. HCl , 68050-08-8; 63, 68050-34-0; 63 (ethyl ester), 68050-33-9; 64, 113508-21-7; 64 (ethyl ester), 113508-43-3; 65, 113508-22-8; 65 (ethyl ester), 113508-44-4; 66, 113508-55-7; 67, 113508-23-9; 68, 113508-56-8; 69, 113508-24-0; 70, 113508-57-9; 71, 113508-25-1; 72, 113508-26-2; 72 (ethyl ester), 113508-45-5; 73, 113508-27-3; 73 (ethyl ester), 113508-46-6; 74, 113508-28-4; 75, 113508-58-0; 76, 113508-29-5; 77, 113508-59-1; 78, 113508-30-8; 79, 68050-12-4; 80, 68050-45-3; 81, 68050-27-1; 81 (ethyl ester), 68050-26-0; 82,

[^6]113508-31-9; 82 (ethyl ester), 113508-47-7; 83, 113508-32-0; 83 (ethyl ester), 113508-48-8; 84, 113508-60-4; 85, 113508-33-1; 86 113508-61-5; 87, 113508-34-2; 88, 113508-35-3; 88 (ethyl ester), 113508-62-6; 89, 113508-36-4; 90, 113508-37-5; 90 (ethyl ester) 113508-49-9; 91, 76577-59-8; 92, 76577-60-1; 92 (oxime), 76577-61-2 93, 76577-63-4; 94, 76577-88-3; 95, 113508-38-6; 95. HCl, 76577-64-5 96, 76577-87-2; 97, 68050-32-8; 97 (ethyl ester), 68050-31-7; 98 113508-39-7; 98 (ethyl ester), 113508-50-2; 99, 68050-36-2; 99 (ethyl ester), 68050-35-1; 100, 68050-29-3; 101, 76577-54-3; 102, 76577 55-4; 102 (oxime), 76577-56-5; 103, 76577-58-7; 104, 113508-40-0; 104 (ethyl ester), 113508-51-3; 105, 68050-25-9; 105 (ethyl ester), 68050-24-8; 106, 68050-16-8; 106 (ethyl ester), 68050-15-7; 107, 68050-19-1; 107 (ethyl ester), 68050-18-0; 108, 113508-41-1; 108 (ethyl ester), 113508-52-4; 109, 68050-40-8; 109 (ethyl ester), 68050-39-5; 110, 68050-30-6; 110 (ethyl ester), 68050-29-3; 111, 68050-21-5; 112, 68050-46-4; 112 (ethyl ester), 68050-21-5; 113 113508-42-2; 113 (ethyl ester), 113508-53-5; 114, 76002-75-0; 114 (ethyl ester), 76013-27-9; 115, 76002-76-1; 116, 76002-77-2; 117, 76002-86-3; 118, 76577-47-4; 119, 76577-41-8; 120, 76577-43-0; 121, 76577-42-9; 122, 76577-46-3; 123, 76002-83-0; 124, 76325-62-7; 125 76325-51-4; 126, 113508-65-9; 127, 76325-47-8; 128, 76325-49-0; 129, 76325-50-3; 130, 76325-52-5; 131, 76325-54-7; 132, 76325-56-9; 133, $76325-55-8$; 134, $76577-90-7$; $134 \cdot \mathrm{HCl}, 76577-80-5$; 135 , $76577-65-6 ; 136,113508-66-0 ; 137,113508-67-1 ; 138,113508-68-2$; 140, 69015-16-3; 142, 69015-14-1; 143, 69015-25-4; 144, 69015-24-3 145, 69015-32-3; 146, 69015-34-5; 147, 69015-35-6; 147 (ethyl ester) 69040-91-1; 148, 69015-36-7; 149, 69015-37-8; 149 (ethyl ester) 69015-26-5; 150, 69015-33-4; 150 (ethyl ester), 69015-23-2; 151 69015-40-3; 151 (ethyl ester), 69015-29-8; 152, 69015-38-9; 152 (ethyl ester), 69015-27-6; 153, 69015-39-0; 153 (ethyl ester) 69015-28-7; 154, 113508-69-3; 154 (ethyl ester), 113508-70-6; 155, 69015-42-5; 155 (ethyl ester), 69015-31-2; 156, 69015-54-9; 156 (ethyl ester), 69015-53-8; 157, 69015-41-4; 157 (ethyl ester), 69015-30-1; 158, 76577-69-0; 159, 76577-70-3; 160, 76577-73-6; 161 76577-74-7; 162, 113508-75-1; 163, 113532-99-3; 165, 99867-03-5 $166,113508-71-7 ; 167,113508-76-2$; 168, 113508-77-3; 169 113508-78-4; 170, 113508-72-8; 171, 113508-73-9; 172, 113508-74-0;
$173,113508-79-5 ; 174,113508-80-8 ; 175,113508-81-9 ; 176$, 82296-39-7; 177, 82296-41-1; 178, 90510-54-6; 179, 113508-82-0; 180, 82296-43-3; 181, 82296-45-5; 182, 65565-70-0; 183, 113508-83-1; $184,113508-85-3 ; 185,113508-84-2 ; 186,38922-77-9 ; 186 \cdot \mathrm{HBr}$, 2549-17-9; 187, 76577-82-7; 188, 76577-83-8; 190, 77947-41-2; 191, 88220-27-3; 192, 60050-77-3; 193, 113508-91-1; 194, 113533-00-9; $195,113508-86-4$; 196, 113508-92-2; 197, 113508-87-5; 198, 64951-05-9; 199, 64951-09-3; 200, 81021-97-8; 201, 113508-88-6; 202, 113508-93-3; 203, 113508-89-7; 204, 113508-94-4; 205, 113508-90-0; $\mathrm{H}_{2} \mathrm{C}=\mathrm{CHCN}, 107-13-1 ; \mathrm{BrCH}_{2} \mathrm{COCO}_{2} \mathrm{Et}, 10-23-1$; $\mathrm{H}_{3} \mathrm{CCHBrCOCO}_{2} \mathrm{Et}, 57332-84-0 ; \mathrm{BrCH}_{2} \mathrm{CO}_{2} \mathrm{Et}, 105-36-2 ; \mathrm{H}_{0} \mathrm{CC}-$ (OEt) ${ }_{3}$, 78-39-7; 5-aminotetrazole, 4418-61-5; 2-amino-4-chloroquinoline, 20151-42-2; 2-amino-4-(methylthioquinoline), 113508-10-4; 2-amino-4-(methylsulfinyl)quinoline, 113508-11-5; ethyl 2-aminoquinoline-3-carboxylate, 36926-83-7; 2-amino-3(hydroxymethyl)quinoline, 75353-55-8; 2-chloro-7-(trifluoromethyl)quinoline, 83183-56-6; 2-amino-7-(trifluoromethyl)quinoline, 113508-12-6; 2-chloro-6-nitroaniline, 769-11-9; 2-chloro-6-nitrodihydrocinnamonitrile, 113508-13-7; 2-amino-5chloroqumoline, 68050-37-3; quinoline, 91-22-5; 2-aminoquinoline, 580-22-3; 2-amino-7-chloroquinoline, 43200-95-9; 2-amino-4.7dichloroquinoline, 68050-28-2; 2-amino-4-chloro-7-methoxyquinoline, 68050-20-4; glycerolacetonide, 100-79-8; 2-amino-6-chloro-4-phenylquinoline, 51478-40-1; 4,5-dichloro-o-phenylenediamine, 5348-42-5; alloxan hydrate, 3237-50-1; dichloroalloxazine, 58590-56-0; 2-amino-6,7-dichloroquinoxaline, 76002-68-1; 2aminoquinoxaline, 5424-05-5; 2-aminoquinoxaline-3-carboxamide, 67568-30-3; 2-amino-3-chloroquinoxaline, 34117-90-3; 2 -amino3 -chloroquinoxalinonium bromide, $76002-73-8$; 2,3,6,7-tetrachloroquinoxaline, 25983-14-6; 2-amino-6,7-dichloro-3-ethoxyquinoxaline, 76325-57-0; 2-amino-3-ethylquinazozlin-4(3H)-one, 2161-26-4; 1-ethylbenzimidazole, 7035-68-9; ethyl propiolate, 623-47-2; 1-methylbenzimidazole, 1632-83-3; 1-butylbenzimidazole, 4886-30-0; quinaldine, 91-63-4; 2-aminobenzothiazole, 136-95-8; 2-amino-6-methoxybenzothiazole, 1747-60-0; 2-amino-5,6-dimethoxybenzothiazole, 29927-08-0; 2-amino-4-methoxybenzothiazole, 5464-79-9

# 6-(Alkylamino)-3-aryl-1,2,4-triazolo[3,4-a]phthalazines. A New Class of Benzodiazepine Receptor Ligands ${ }^{\dagger}$ 

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> Some 6-(alkylamino)-3-aryl-1,2,4-triazolo[3,4-a]phthalazines have been shown to displace diazepam from rat brain specific binding sites, in vitro, with $K_{\mathrm{i}}(\mathrm{nM})$ values comparable to those of reference benzodiazepines and to have anticonvulsant (pentylenetetrazole test, mice) and anticonflict activity (Vogel test, rat) in vivo. Separation between the doses causing anticonflict effects (Vogel test, rat) and those impairing motor coordination (rotarod test, rat) has been shown for $N, N$-bis(2-methoxyethyl)-3-(4-methoxyphenyl)-1,2,4-triazolo[3,4-a]phthalazin-6-amine (80). This compound, unlike diazepam, was inactive in counteracting the strychnine (mouse) and maximal electroshock (mouse) induced convulsions and in the "aggressive monkey" model. These differences from the classical benzodiazepines in the animal tests indicate that $\mathbf{8 0}$ may have some selective anxiolytic activity.

The search for antianxiety agents without nonspecific central nervous system (CNS) depressant side effects has led to the discovery of several classes of compounds chemically unrelated to the benzodiazepines (BZ). The

[^7]field has been recently reviewed, ${ }^{1-5}$ and since then a few more examples of this type of compounds have been reported. ${ }^{6-10}$ All these compounds share the property of

[^8]
[^0]:    (1) Patterson, R. In Current Perspectives in Allergy; Goetzl, E. J., Kay, A. B., Eds.; Churchill Livingstone: New York, 1982; p 17.
    (2) Cox, J. S. G.; Beach, J. E.; Blair, A. M. J. N.; Clarke, A. J.; King, J.; Lee, T. B.; Loveday, D. E. E.; Moss, G. F.; Orr, T. S. C.; Richie, J. T.; Sheard, P. Adv. Drug. Res. 1970, 5, 115.
    (3) Turner-Warwick, M. in ref 1, p 153.
    (4) Bernstein, I. L. J. Allergy Clin. Immunol. 1981, 68, 247.
    (5) Church, M. K. Med. Actual/Drugs Today 1978, 14, 281.

[^1]:    (6) Barnes, A. C.; Hairsine, P. W.; Matharu, S. S.; Ramm, P. J.; Taylor, J. B. J. Med. Chem. 1979, 22, 418.
    (7) Miller, P.; James, G. W. L. Arch. Int. Pharmacodyn. 1978, 231, 328.
    (8) Proposed INN, WHO Chron. 1980, 34 (Suppl. to 9), 21.
    (9) Danswan, G. W.; Hairsine, P. W.; Rowlands, D. A.; Taylor, J. B.; Westwood, R. J. Chem. Soc., Perkin Trans. 1 1982, 1049.

[^2]:    (18) Evdokimoff, V. Rend. Ist. Super. Sanita (Engl. Ed.) 1960, 23, 542.
    (19) Freri, M. Gazz. Chim. Ital. 1938, 68, 616.
    (20) Chapman, D. D. J. Org. Chem. 1972, 37, 2498.
    (21) Biekert, E.; Koessel, H. Justus Leibigs. Ann. Chem. 1963, 662, 83.

[^3]:    (26) Kadin, S. B. U.S. Patent 407534321 Feb. 1978.

[^4]:    (27) Ager, I. R.; Ramm, P. J. U.K. Patent 1596 652, 20 Jan. 1977.
    (28) Barnes, A. C.; Ramm, P. J. U.K. Patent 2027707, 31 July 1979.
    (29) Miller, P., unpublished results.

[^5]:    (40) Pettit, G. W.; Neill, A. B. Can. J. Chem. 1964, 42(7), 1764. (41) Dandegaonker, S. H.; Revanker, G. R. J. Karnatak Univ. 1961, $6,25$.

[^6]:    (47) Stoss, P.; Satzinger, G. Chem. Ber. 1976, 109, 2097.
    (48) Gupta, R. R.; Ojha, K. G.; Kumar, M. J. Heterocycl. Chem. 1980, 17, 1325.

[^7]:    ${ }^{\dagger}$ This work is dedicated to Prof. Valdo Mazzi, Institute of Comparative Anatomy, University of Torino (Italy) on occasion of his 70th birthday.
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[^8]:    (1) Williams, M. J. Med. Chem. 1983, 26, 619.
    (2) Goldberg, M. E.; Salama, A. I.; Patel, J. B.; Malick, J. B. Neuropharmacology 1983, 22, 1499.
    (3) Martin, I. L. Trends Pharmacol. Sci. 1984, 5, 343.
    (4) Williams, N.; Yokoyama, N. Annu. Rep. Med. Chem. 1986, 21, 11.
    (5) Haefely, W.; Kyburz, E.; Gerecke, M.; Mohler, H. In Advances in Drug Research; Testa, B., Ed.; Academic: New York, 1985; Vol. 14, pp 166-322.
    (6) Guzman, F.; Cain, M.; Larscheid, P.; Hagen, T.; Cook, J. M.; Schweri, M.; Sholnick, P.; Paul, S. J. Med. Chem. 1984, 27, 564.

